#### Ion Beam Interactions with Advanced Photoresist Polymers

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### Introduction

- Motivation:
  - There is little understanding of fundamental etching and roughening mechanisms of polymer masking materials.
  - Increased etch resistance is required as thinner masking films are needed.
  - Photoresist roughening, resulting in poor pattern transfer, becomes increasingly important as device dimensions shrink.
- Goals
  - Determine important variables when considering roughness and etching mechanisms of polymer masking materials.



- Outline: Beam system studies
  - Sputtering of photoresist under ion bombardment.
  - Roughening of photoresist under ion bombardment.

E. Hudson, Z. Dai, et al. "Control of Line Edge Roughness for Etching with 193nm Photoresist." *Proc. Dry Process Int. Symp.* 253 (2003).

#### **Experimental Technique**

- UHV Chamber:
  - Base Pressure: ~5x10<sup>-8</sup> Torr pumped with a 2000 L•s<sup>-1</sup> turbo pump
- PHI Model 04-191 Ion Gun:
  - Chamber pressure rises to ~1x10<sup>-6</sup> Torr
  - Ions: He+, Ar+ and Xe+
  - Energies: 0.5 keV and 1 keV
  - Beam Size: ~0.5 cm
  - Substrate temperature control
  - Neutralizing filament to prevent surface charging



### Samples and Analytical Tools

- Samples:
  - Rohm and Haas methacrylate-based 193 nm photoresist
    - Coated on Quartz Crystal Microbalances (QCM) for sputter yield studies
    - Samples cleaved from coated 8" wafers for roughness studies
- Analytical Tools:
  - Sputter yield studies: QCM monitored for mass change during ion impact
  - Roughness studies: ex situ Atomic Force Microscopy (AFM) surface roughness analysis

- Ion Flux  $\Phi$  (ions-cm<sup>-2</sup>-s<sup>-1</sup>) on the sample:
  - I = current measured by Faraday cup (~nA)
  - $A = Faraday cup area (A = 4.55x10^{-4} cm^{-2})$
  - q = elemental charge
- Ion Fluence  $\Gamma$  (ions-cm<sup>-2</sup>):
  - $-\Phi = \text{ion flux (ions \cdot cm^{-2} \cdot s^{-1})}$
  - t = exposure time (s)

$$\Gamma = \Phi \cdot t$$

- Equivalent carbon etch yield (EY) (eq C-ion<sup>-1</sup>):
  - Number of equivalent carbons removed per incoming ion
  - Obtained from slope of mass removed vs. fluence plot from QCM measurements.

$$\Phi = \frac{I}{A \cdot q}$$



### Steady-state etch yield:

#### comparison to plasma experiments

Rohm and Haas 193 nm photoresist

- Ion beam etch yields:
  - 0.5 and 1 keV ion beam steady-state etch yields from QCM experiments for fluences > 5x10<sup>16</sup> ions•cm<sup>-2</sup>.
- Plasma etch yields:
  - Argon plasma experiment etch yields



# Evolution of etch yield with fluence: comparison of 193 nm and 248 nm photoresist



- More mass removed prior to reaching steady-state for 193 nm photoresist.
- Comparison of steady-state etch yields:
  - Higher ion energy results in higher etch yields (for a given material).
  - Etch yield of 193 nm is higher than 248 nm (at a given energy).

#### Comparison of steady-state etch yields

• Empirical formula: etch yield is proportional to Ohnishi parameter<sup>1</sup>





<sup>1</sup>Gokan, H., S. Echo, and Y. Ohnishi, Journal of the Electrochemical Society, 1983. **130**(1): p. 143-146. <sup>2</sup>National Physics Laboratory, UK

### Surface roughening of 193 nm photoresist: Argon ion bombardment

- Surface roughness obtained from 1x1  $\mu$ m<sup>2</sup> AFM images.
- Ar<sup>+</sup> bombardment: ion energy and substrate temperature
- Surface roughness of unprocessed sample: ~0.3 nm



- Decreasing ion energy increases roughness.
- Increasing substrate temperature increases roughness.
- Heating alone does not roughen the surface.

#### Surface roughening of 193 nm photoresist: effect of Ar<sup>+</sup> ion energy

Ion energy effect (20°C):
0.5 keV



1.27 nm



0.30 nm

- Ion beam studies: increased roughness at lower ion energies
- Also seen on polystyrene derived materials.
- Argon plasma exposures at lower ion energy resulted in greater roughness.

### Surface roughening of 193 nm photoresist: effect of substrate temperature

 Substrate temperature effect (0.5 keV Ar<sup>+</sup>): 20°C





#### 1.27 nm



45°C

- Literature: increasing substrate temperature on amorphous materials should promote diffusion, decreasing surface roughness.
- Uniqueness of polymers:
  - highly modified surface layer
  - large gradient in composition and structure over a few nanometers

#### Surface roughening of 193 nm photoresist: Xenon ion bombardment



- Similar trends to argon bombardment:
  - Lower ion energy produces increased surface roughness
  - Increased substrate temperature produces increased surface roughness
- Roughness is greater than Ar<sup>+</sup> bombardment.
- Roughness develops with 1 keV Xe<sup>+</sup> bombardment.

## Surface roughening of 193 nm photoresist: ion energy and substrate temperature



E. Hudson, Z. Dai, et al. "Co *Int. Symp.* 253 (2003).

#### Surface roughening of 193 nm photoresist: comparison of Xe<sup>+</sup>, Ar<sup>+</sup>, and He<sup>+</sup> bombardment

(fluence ~1.3x10<sup>17</sup> ions•cm<sup>-2</sup> for all samples)



#### Conclusions: sputtering of polymers

- Polymer sputtering characterized by an initial high etch rate. A lower steady-state etch yield similar to that of carbon is reached after fluences of ~5x10<sup>16</sup> ions•cm<sup>-2</sup>.
- Steady-state etch yields of Ar<sup>+</sup> bombardment follow the empirical Ohnishi parameter taking into account inherent chemical effects of the polymer. Ohnishi parameter does not necessarily hold true in the presence of chemistry.
- The amount of material removed prior to reaching steady-state is polymer dependent.
  - more mass removed prior to reaching steady-state for 193 nm photoresist compared to 248 nm photoresist
- Ion beam etch yields consistent with argon plasma experiments.

#### Conclusions: roughening of polymers

- Ion energy effect
  - Increased roughness at lower ion energies (0.5 keV > 1 keV)
- Ion mass effect
  - Increased roughness with increased ion mass ( $Xe^+ > Ar^+ > He^+$ )
- Substrate temperature effect
  - Increased roughness with increased substrate temperature (45°C > 20°C)
- Experiments completed on polystyrene derived materials are consistent with these conclusions.
- Any roughness theory developed that also includes chemical effects must agree with these observations.

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