

Depth Profiling of Organic Photovoltaic and OLED Materials by Cluster Ion Beams

J.S. Hammond¹, S. N. Raman¹, S. Alnabulsi¹, N. C. Erickson² and R. J. Holmes²

1. Physical Electronics, 18725 Lake Drive East, Chanhassen, MN.

****2. University of Minnesota, Minneapolis, MN.***

Challenges for Organic Electronics Research

- ❑ **Efficiencies are based on optimum band matching and the physical dispersions of components**
 - Work functions at discrete interfaces e. g. metal electrodes
 - Diffusion lengths of charge carriers
- ❑ **Device lifetimes influence customer acceptance and profit margins**
 - Atmospheric contamination (H_2O , O_2) degrade chemistry
 - Flexible designs challenge encapsulation technologies
 - Breakdown mechanisms are not well understood
- ❑ **Chemical and molecular specific depth profiling is desired**
 - Depth resolution of a few nm with high sensitivity
 - Cluster ion depth profiling interleaved with XPS and TOF-SIMS may be solution

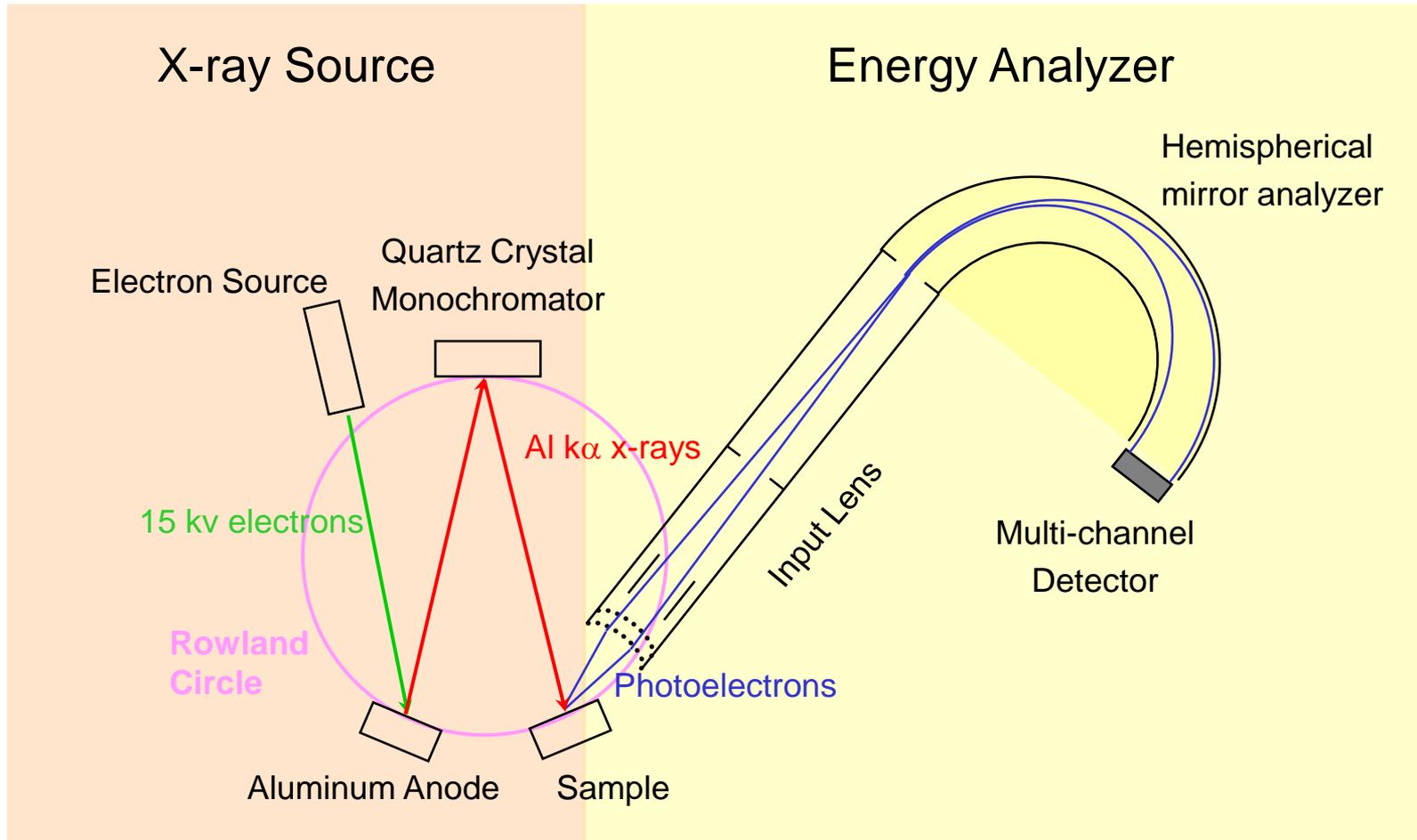
Why use Depth Profiling for Organic Electronics?

- ❑ **New “nanotechnology” products use extremely thin organic and polymer structures**
 - OLEDs
 - Energy conversion materials and fuel cell membranes
- ❑ **Fabrication process producing molecular gradients can result in significant differences in efficiency**
- ❑ **Product degradation can result from molecular oxidation and molecular diffusion**
- ❑ **Spectroscopy with a nano-scale depth of analysis (XPS and TOF-SIMS) needed for surface and depth profiling characterization of molecular composition and diffusion**

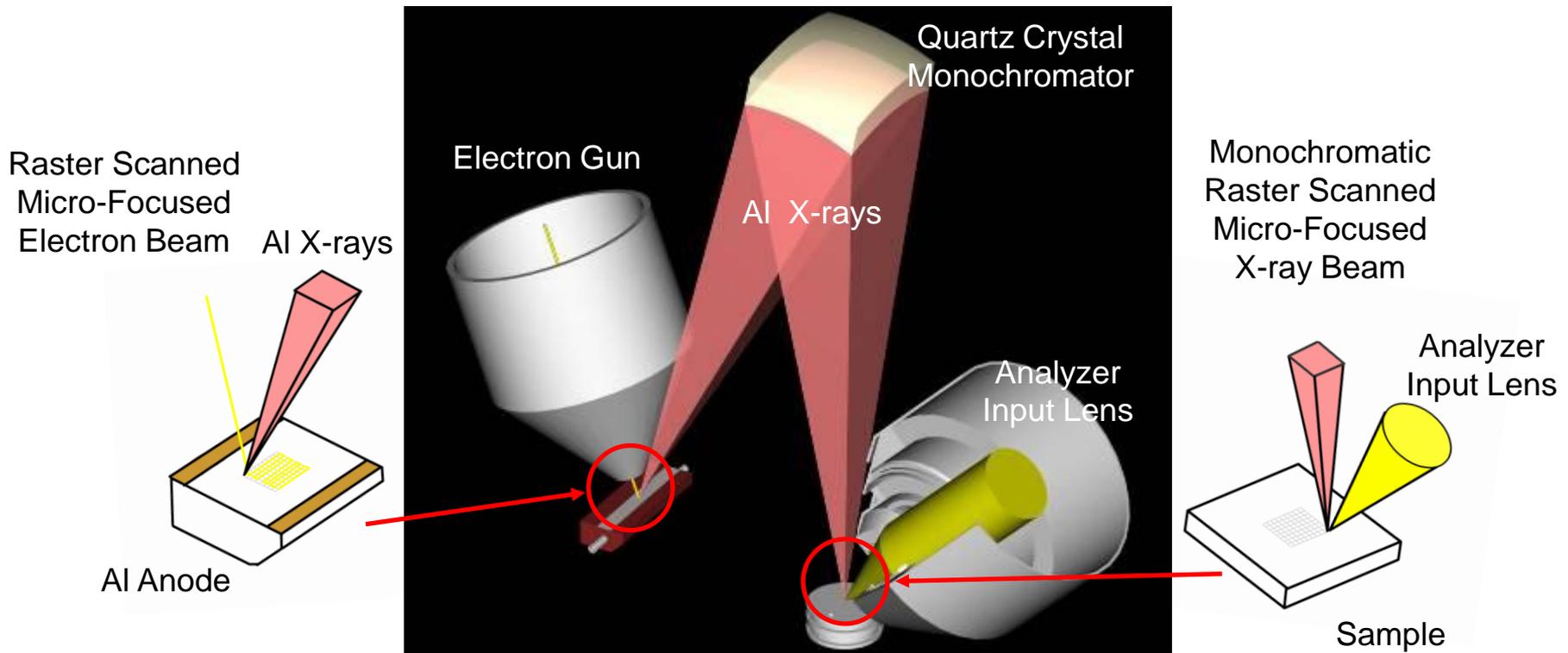
Comparison of XPS and TOF-SIMS

	XPS	TOF-SIMS
Probe Beam	Photons	Ions
Analysis Beam	Electrons	Ions
Spatial Resolution	10 μm	0.10 μm
Sampling Depth(\AA)	5-75	1-10
Detection Limits	0.01atom %	1ppm
Information Content	Elemental Chemical	Elemental Chemical Molecular
Quantification	Excellent	Std. needed

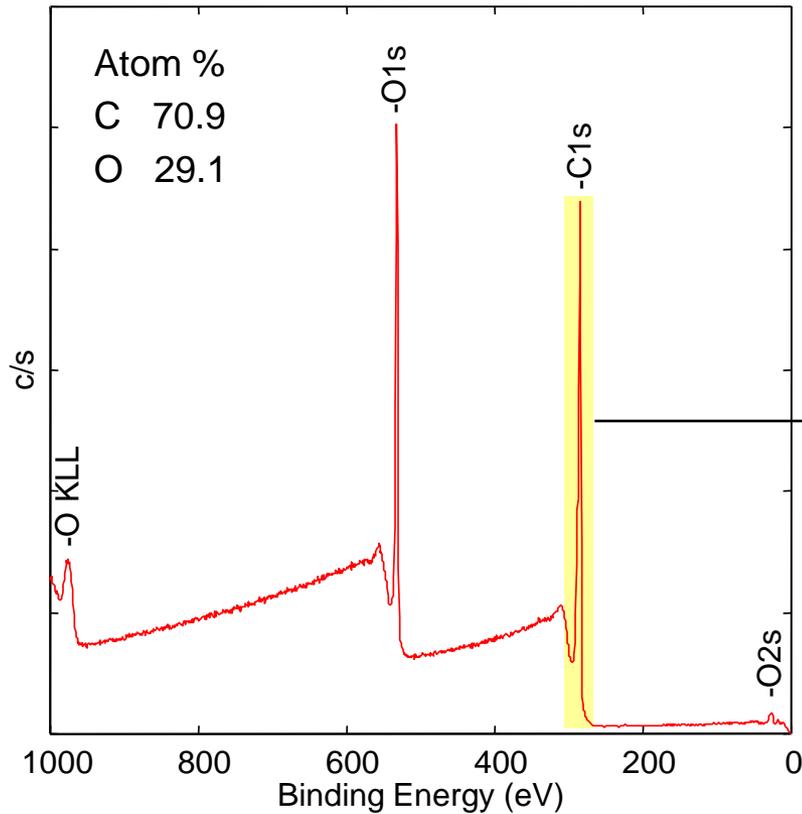
XPS System Schematic



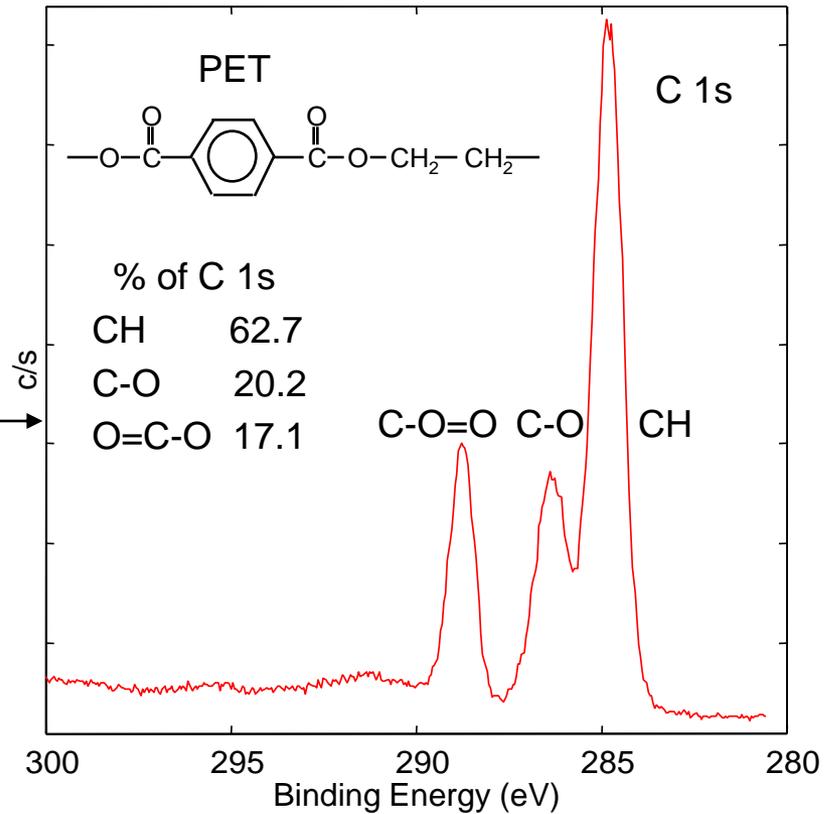
Scanning Micro Focused X-ray Source



Typical XPS Spectra Poly(ethylene terephthalate)



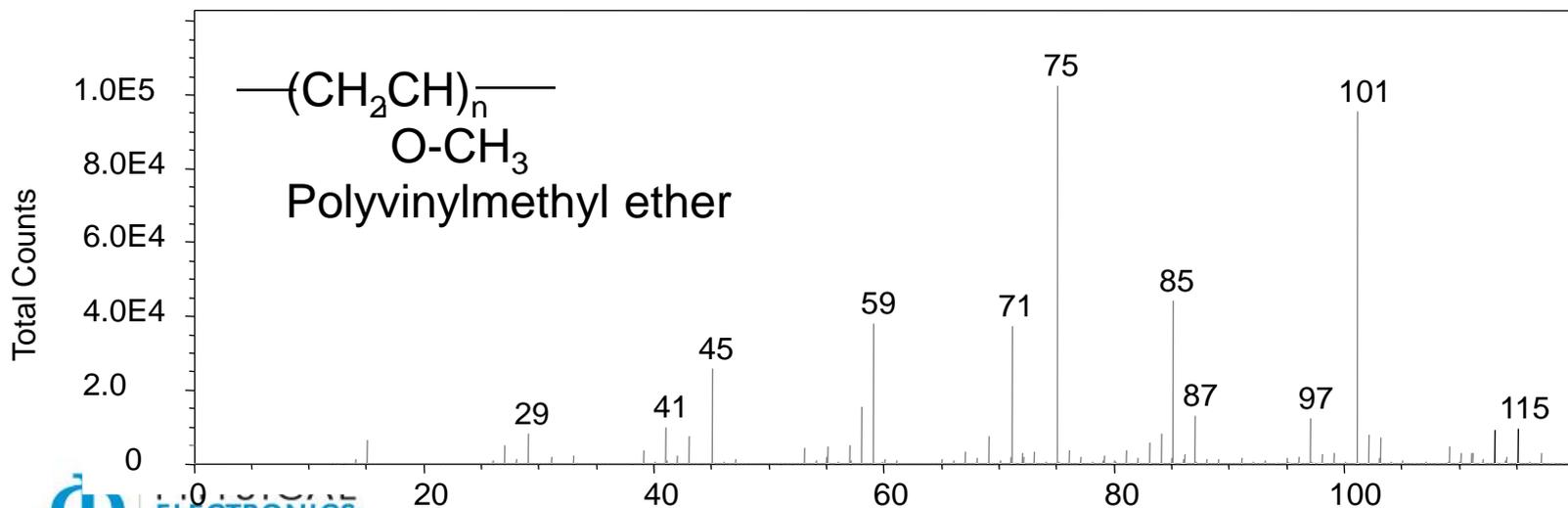
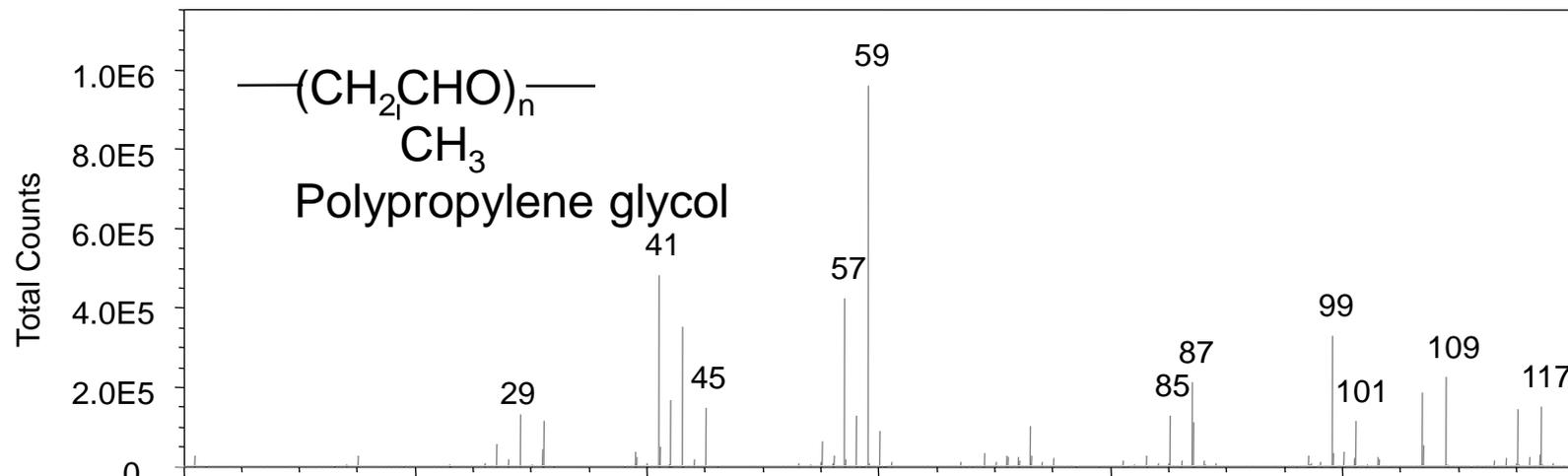
XPS survey spectra provide
quantitative elemental
information



High resolution XPS spectra
provide quantitative chemical state
information

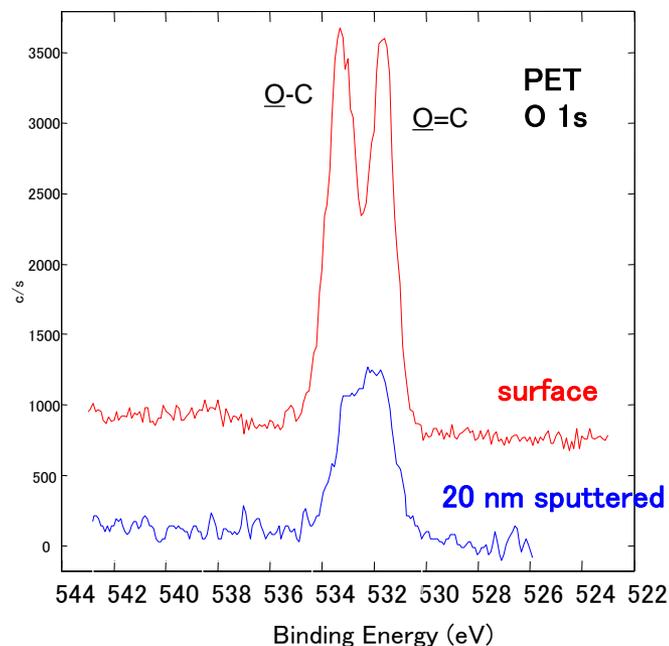
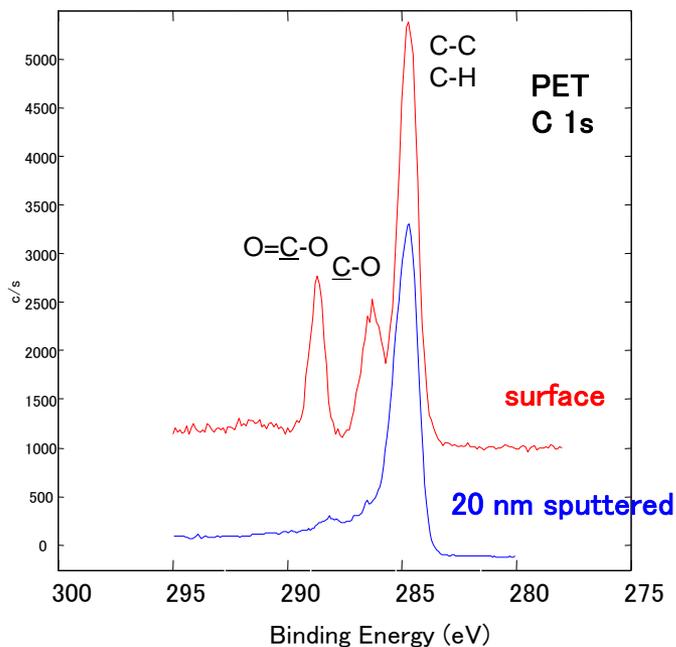
Unique Polymer "Fingerprint" Identification Using TOF-SIMS Spectra

XPS shows identical spectra for both polymers



Damage Accumulation Indicated by XPS

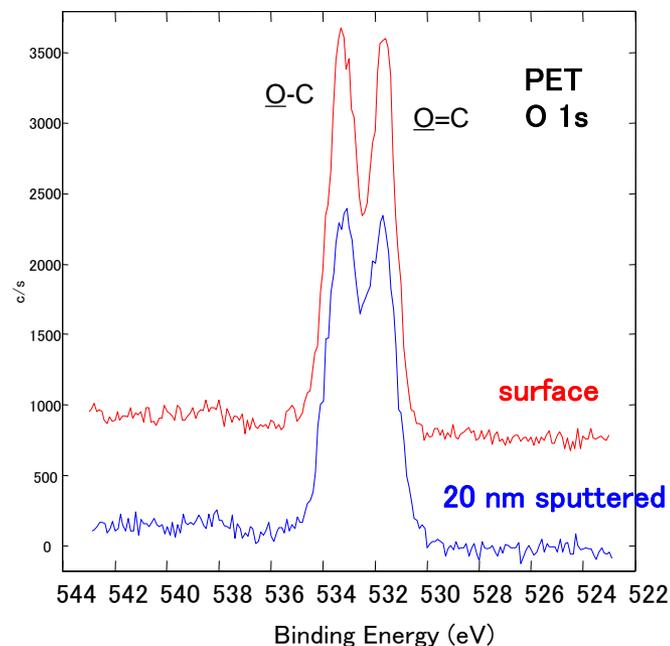
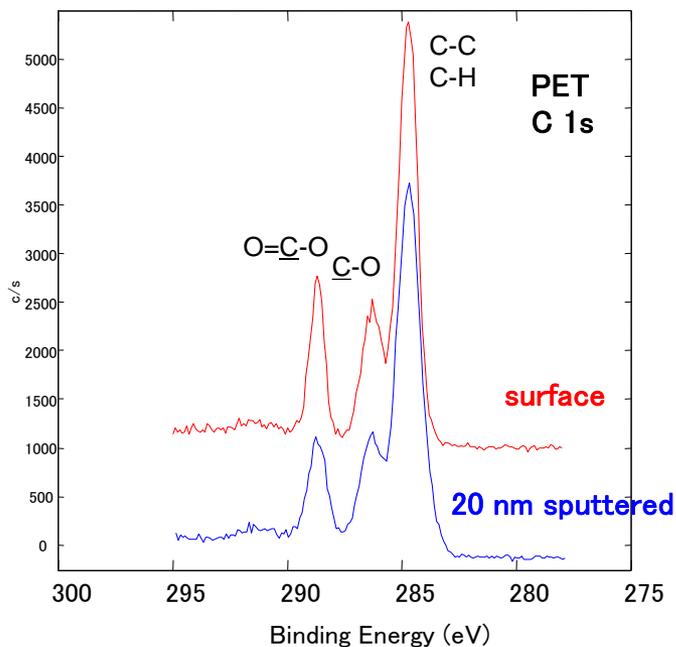
500 eV Ar⁺ Sputter



Damage is observed.

No Damage Observed by XPS

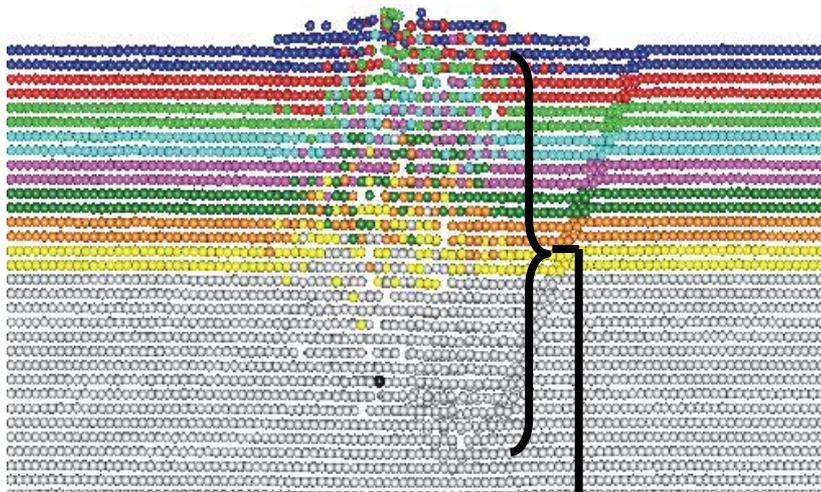
10keV C_{60}^+ Sputter



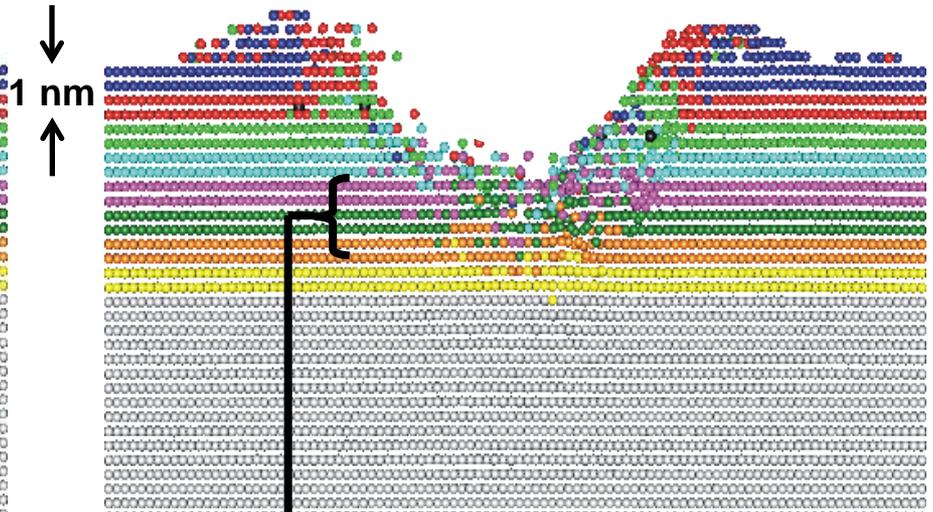
No damage observed.

Altered Volume & Sub-surface Damage

15 keV Ga



15 keV C₆₀

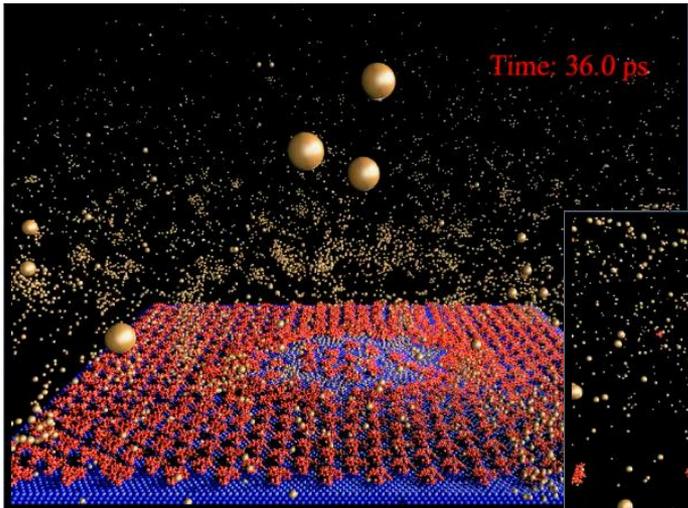


Residual C₆₀ sputter damage is mostly removed with the next C₆₀ impact event.

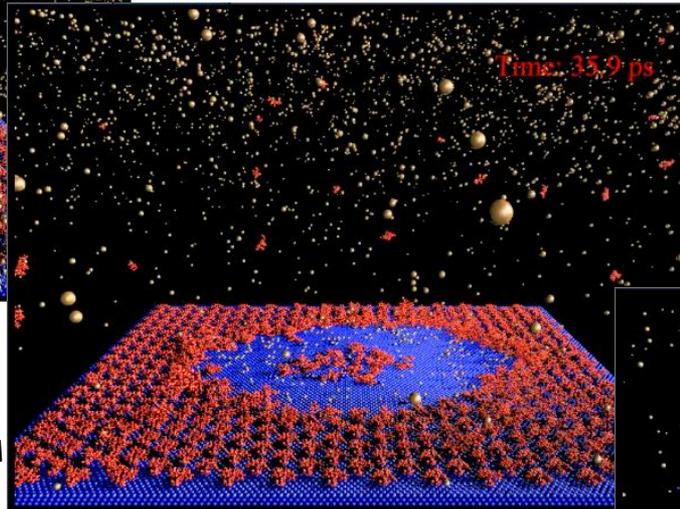
Non-C₆₀ sputter sources result in the **accumulation of sub-surface damage.**

Molecular Dynamics: Ar_{9000}^+ GCIB on PS/Ag

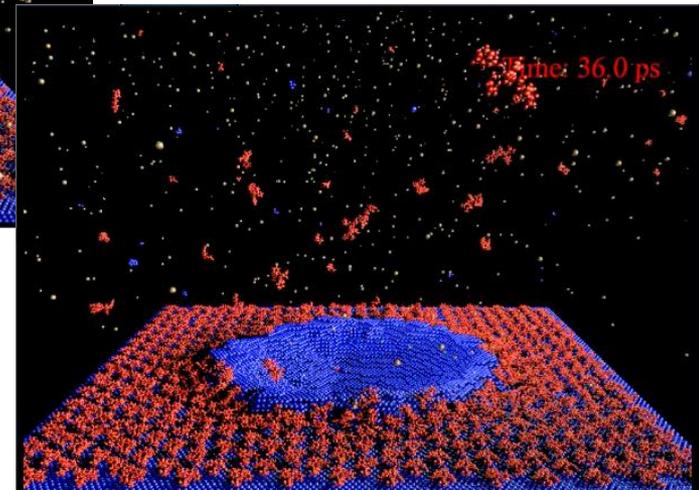
0.5eV/Ar atom



2.0eV/Ar atom



10eV/Ar atom



Lateral Jet Momentum Transfer Sputtering

The permission of the pictures is by courtesy of Professor Zbigniew Postawa, Jagiellonian University (Poland); L. Rzeznik, B. Czerwinski, B.J. Garrison, N. Winograd and Z. Postawa, "Microscopic Insights into the Sputtering of Thin Polystyrene Films on Ag{111} Induced by Large and Slow Ar Clusters", J. Phys. Chem C 112 (2008) 521.

Single Layer, Graded Composition OLED structure

Previous Graded Composition OLED Research

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Highly efficient, single-layer organic light-emitting devices based on a graded-composition emissive layer

Nicholas C. Erickson¹ and Russell J. Holmes^{2,a)}

¹*Department of Electrical and Computer Engineering, University of Minnesota, Minneapolis, Minnesota 55455, USA*

²*Department of Chemical Engineering and Materials Science, University of Minnesota, Minneapolis, Minnesota 55455, USA*

Graded Composition Structures Overview

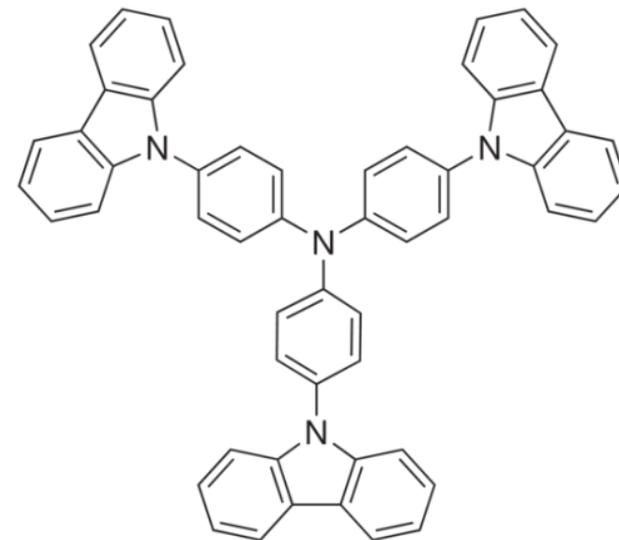
- ❑ Graded Electron and Hole Transport Materials (ETM and HTM) with a green phosphorescent emitter
- ❑ 100% HTM at the Anode and 100% ETM at the Cathode
- ❑ Result
 - At 600 cd/m²
 - Peak external quantum efficiency $\eta_{\text{EQE}} = (19.3 \pm 0.4)\%$
 - Corresponding to internal quantum efficiencies approaching 100%
 - Peak power efficiency $\eta_{\text{p}} = (66.5 \pm 1.3) \text{ lm/W}$
- ❑ These structures are simple to grow
- ❑ But need to confirm the chemistry of these structures is essential

Analytical Technique

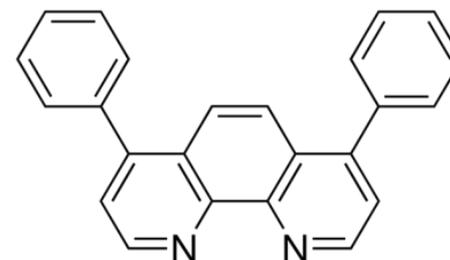
- ❑ **Thin film samples were characterized in a PHI *VersaProbe* II Scanning XPS system equipped with an Ar⁺ Gas Cluster Ion Beam source for depth profiling**
 - The GCIB source can be operated
 - With Beam energies from 2.5 kV to 20 kV
 - Cluster size from <1000 to 5000
- ❑ **The XPS system has excellent charge neutralization capability to compensate for differential charging at various interfaces**

Structure of the two molecules for OLED

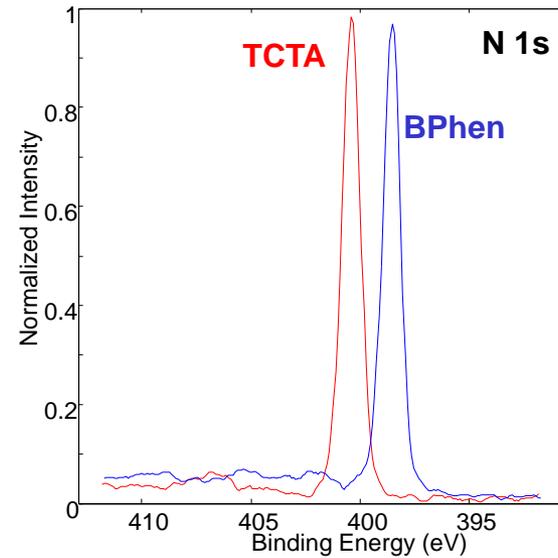
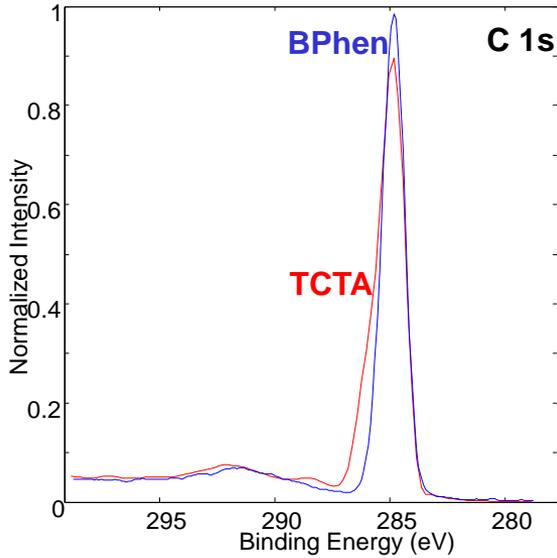
- Tris(4-carbazoyl-9-ylphenyl)amine (TCTA) (Hole Transfer Material)



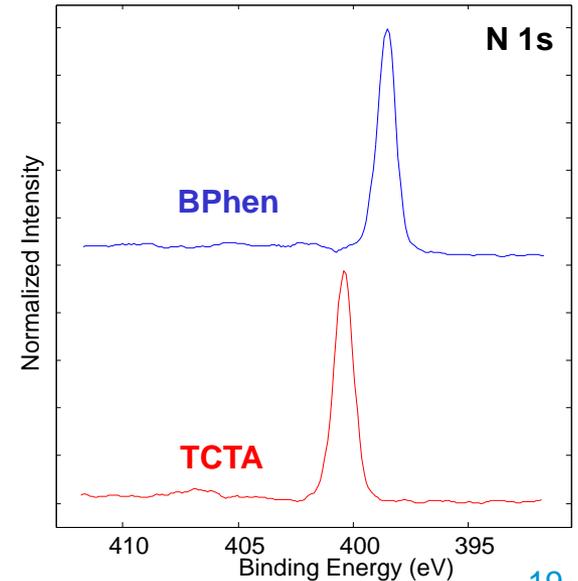
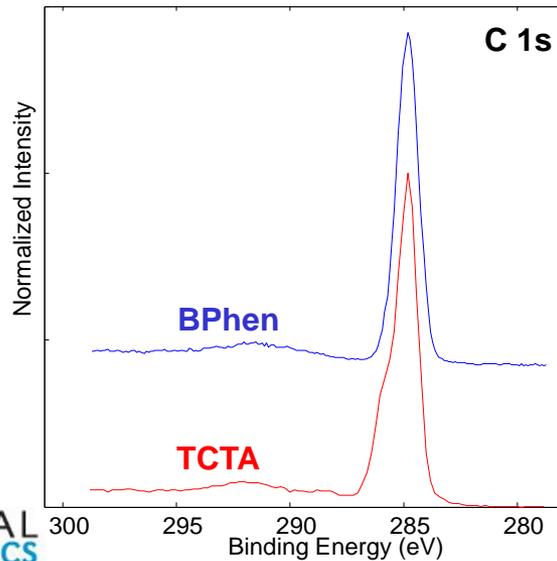
- Bathophenanthroline (BPhen) (Electron Transfer Material)



C 1s and N 1s Spectra for TCTA and BPhen

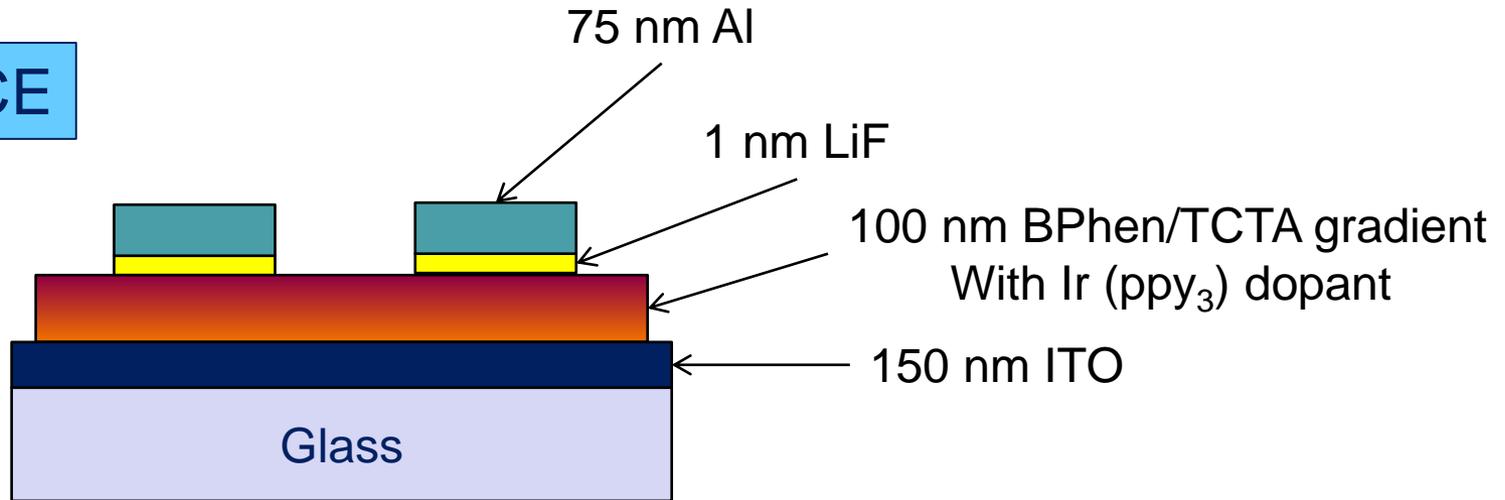


1.8 eV Separation

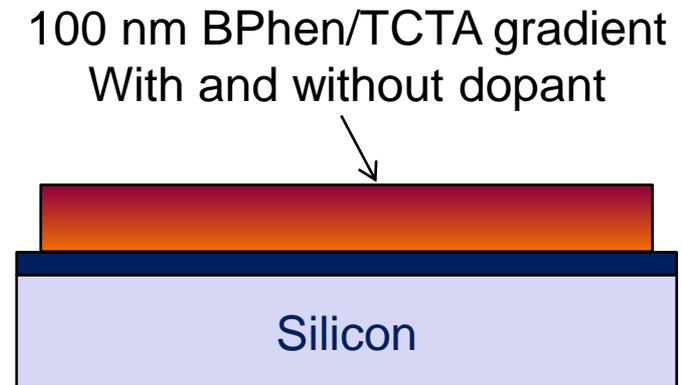


Sample Structure for OLED

DEVICE



XPS TEST STRUCTURES

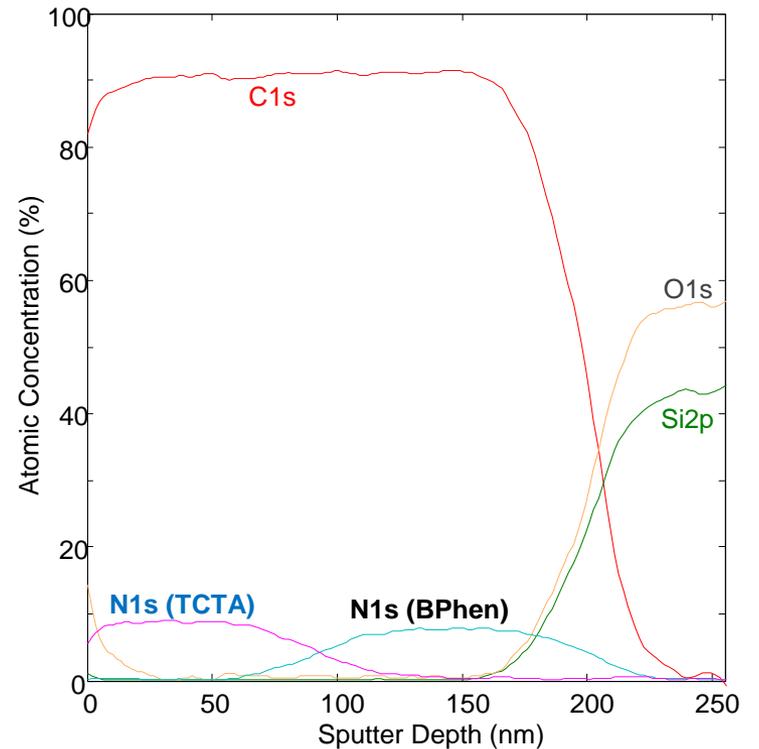
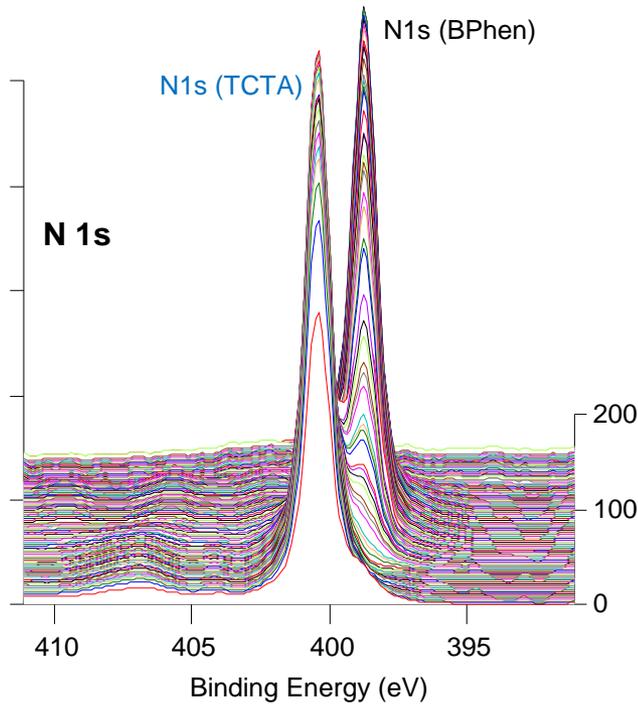


GCIB Depth Profile of TCTA:Bphen/SiO₂/Si



Ar₂₅₀₀, 10 kV, 2 nA, 3 mm x 3 mm

4 eV/atom GCIB

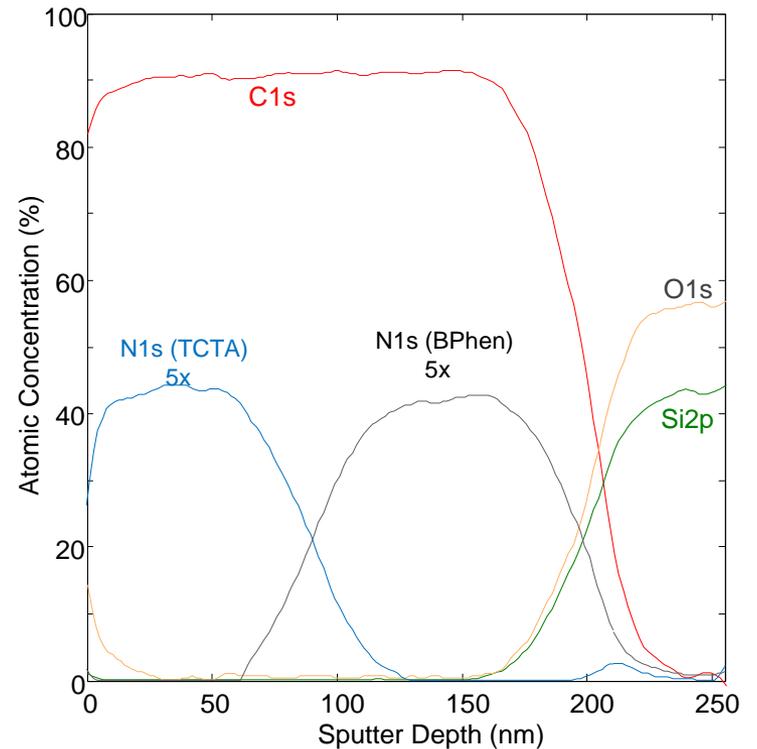
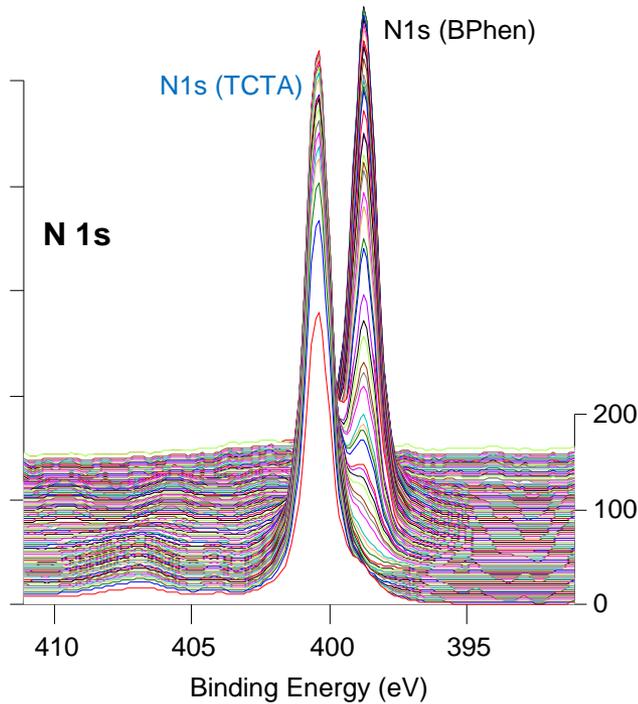


GCIB Depth Profile of TCTA:Bphen/SiO₂/Si

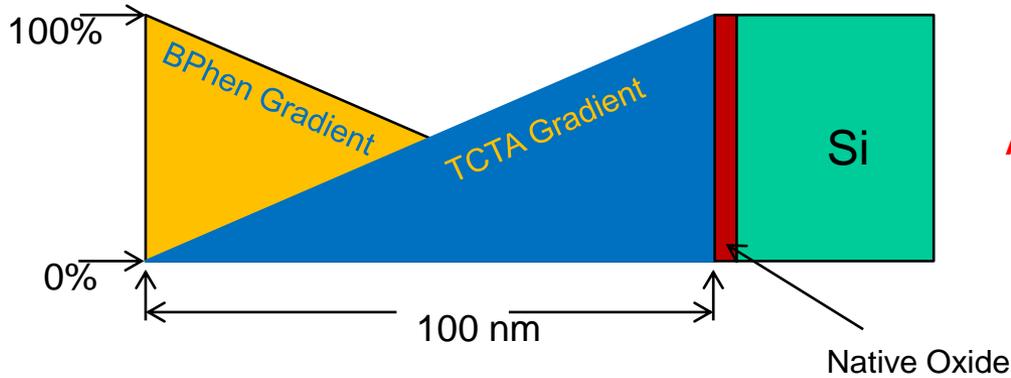


Ar₂₅₀₀, 10 kV, 2 nA, 3 mm x 3 mm

4 eV/atom GCIB

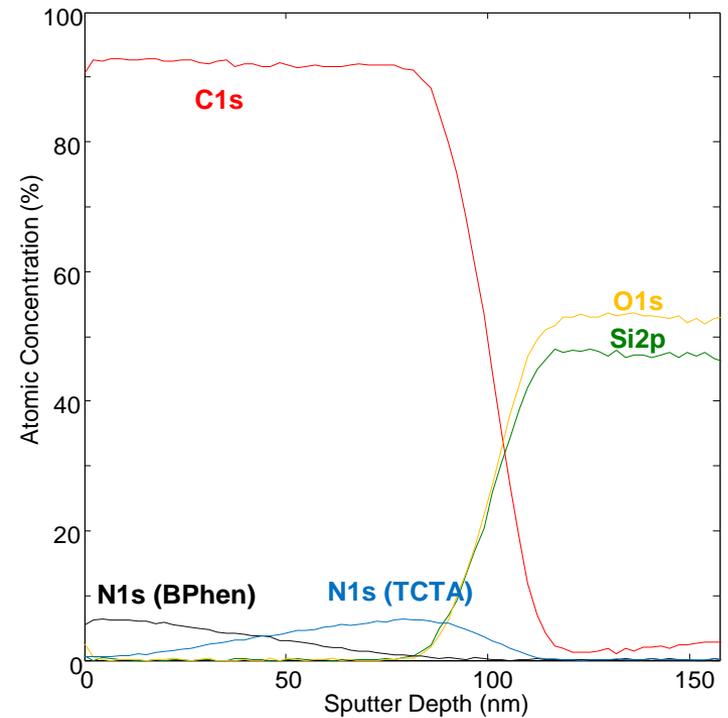
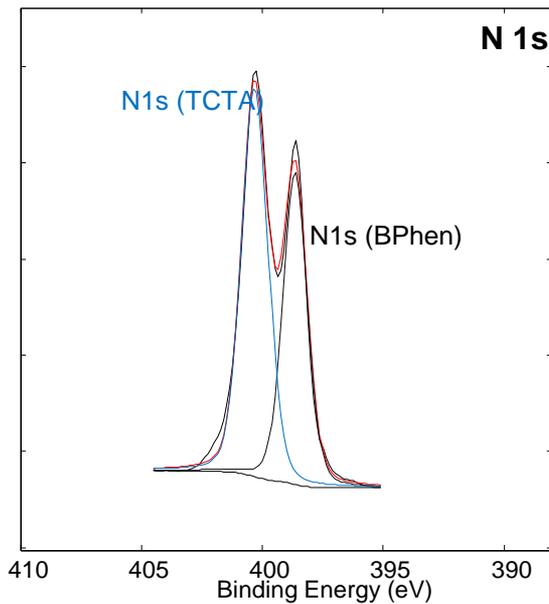


GCIB Depth Profile of Bphen:TCTA/SiO₂/Si

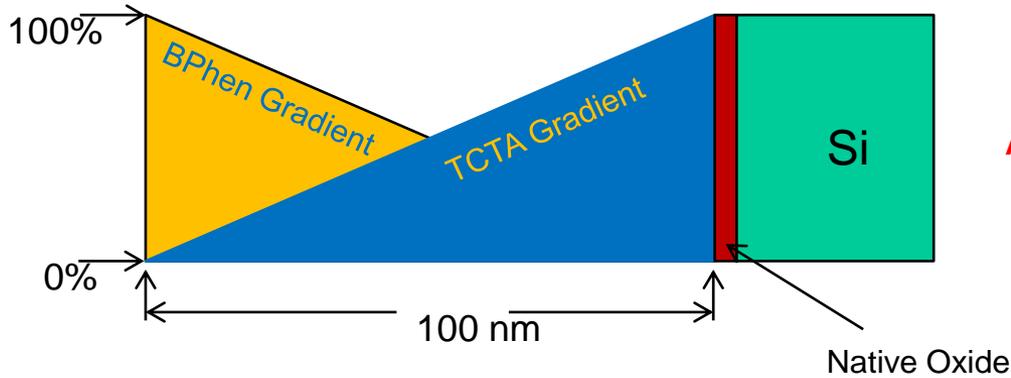


Ar₂₅₀₀, 10 kV, 2 nA, 3 mm x 3 mm

4 eV/atom GCIB

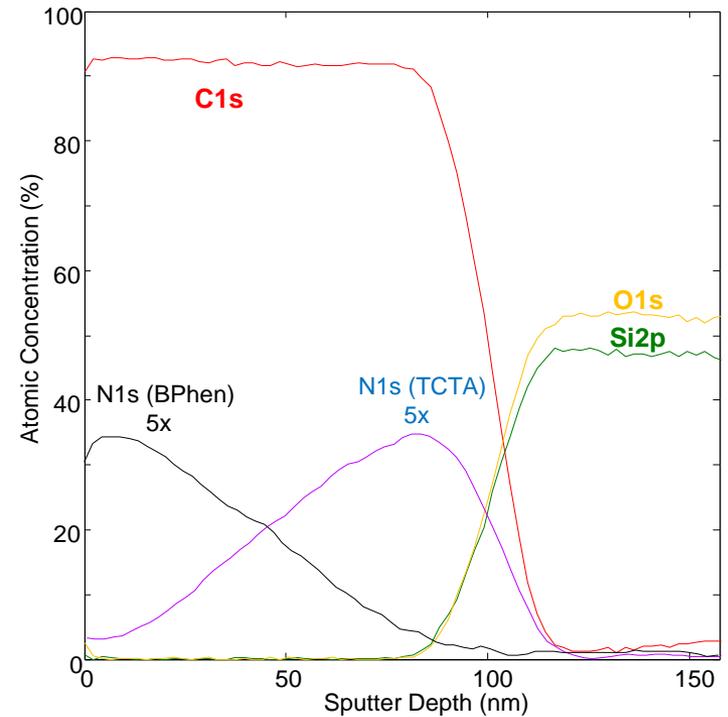
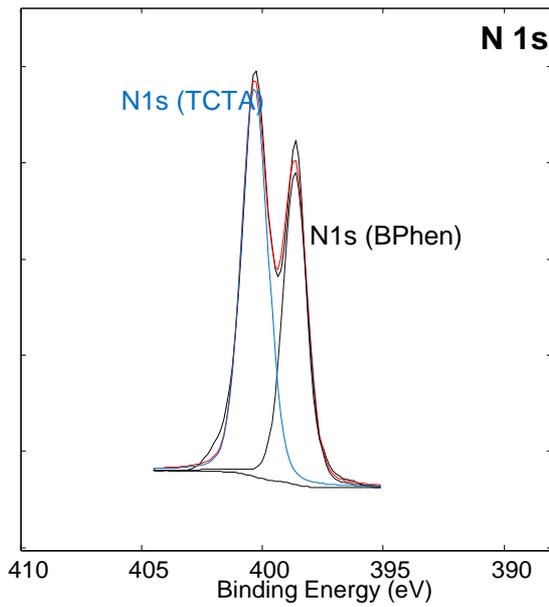


GCIB Depth Profile of Bphen:TCTA/SiO₂/Si

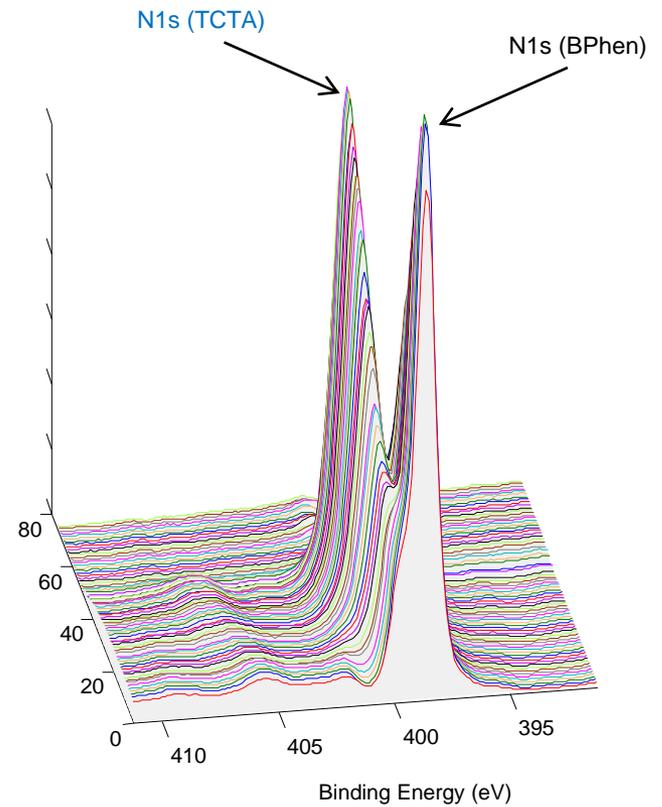
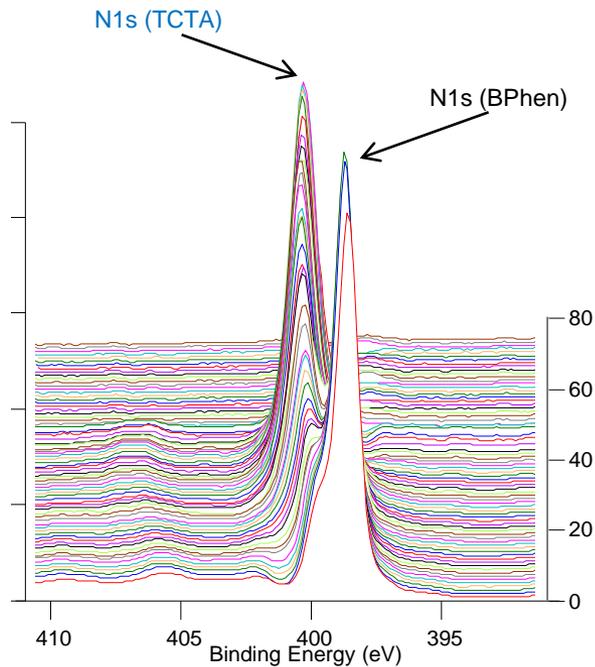
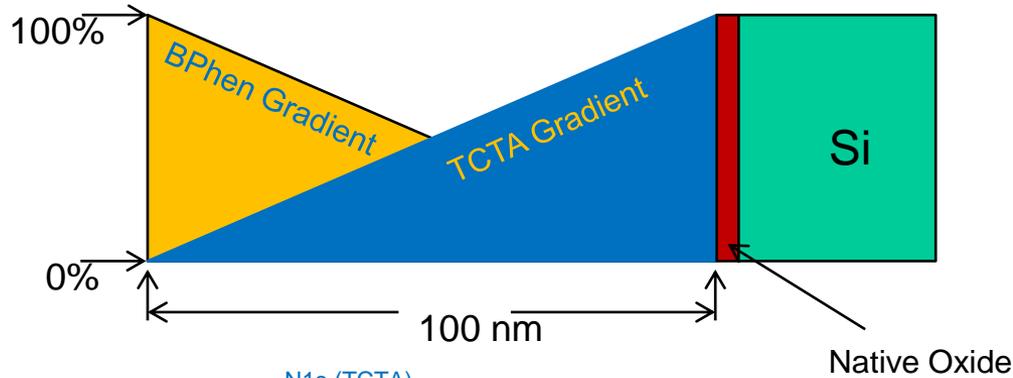


Ar₂₅₀₀, 10 kV, 2 nA, 3 mm x 3 mm

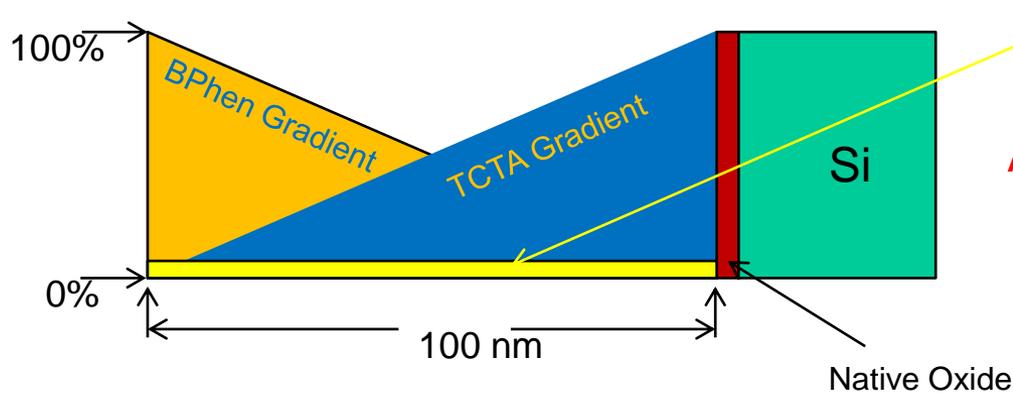
4 eV/atom GCIB



N 1s Spectra from GCIB Depth Profile of Bphen:TCTA/SiO₂/Si



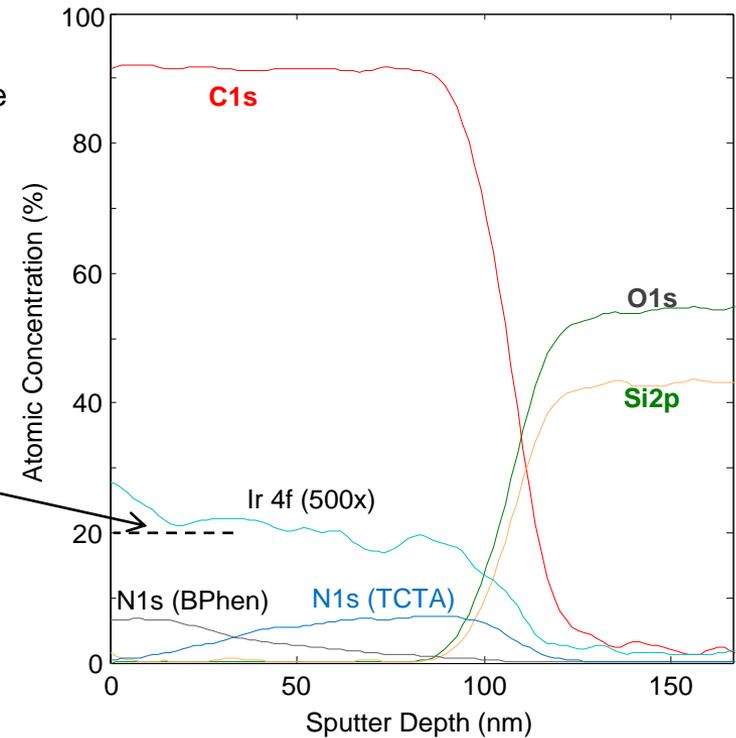
Ir Emissive Compound from GCIB Depth Profile of Bphen:TCTA/SiO₂/Si



Ar₂₅₀₀, 10 kV, 2 nA, 3 mm x 3 mm

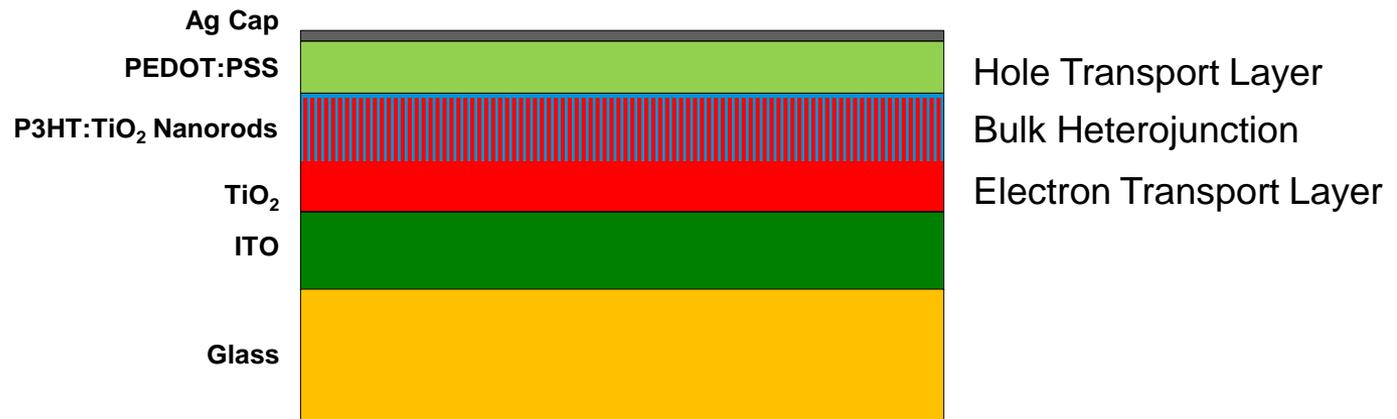
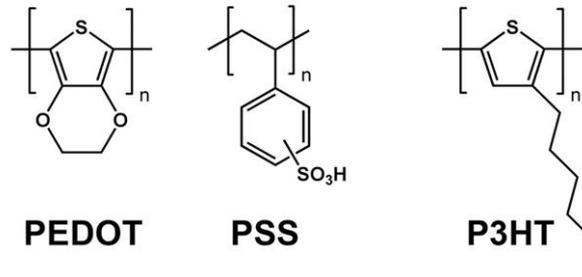
4 eV/atom

Ir 4f concentration 0.04 at %



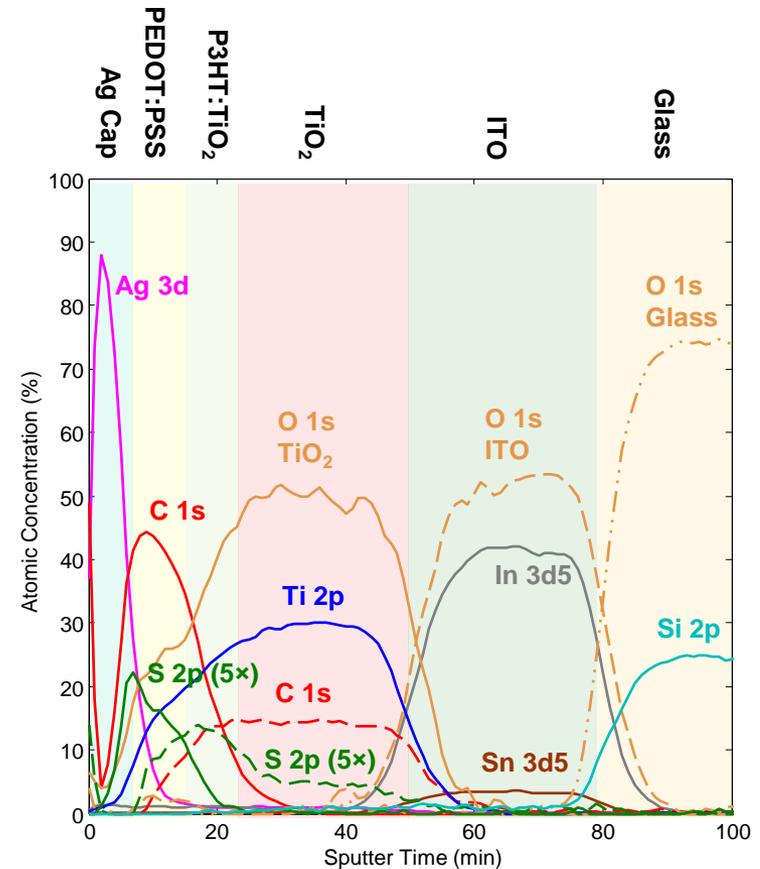
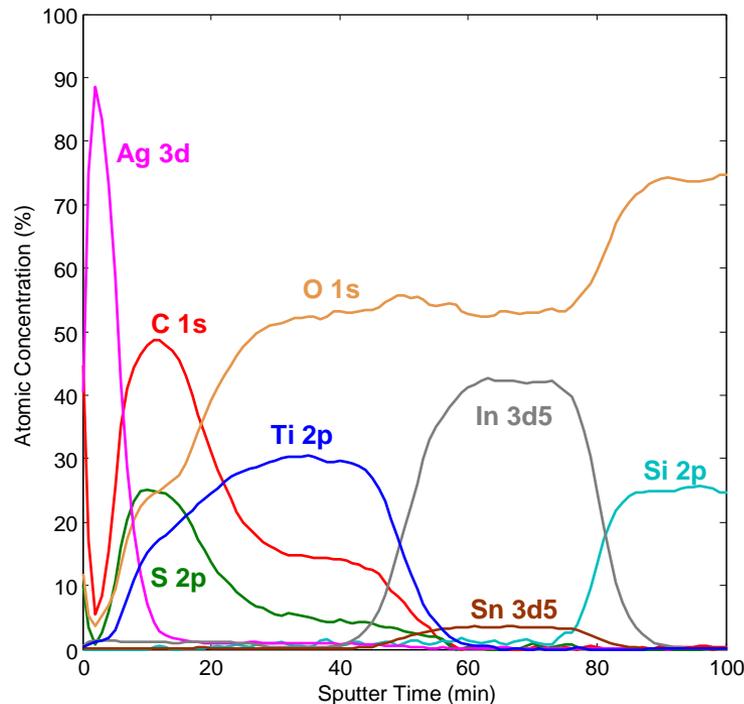
Composite Multi-layer Inverted Organic Photovoltaic Device

Sample is comprised of various material types: Metal / Polymer / Oxide



Composite – Organic Photovoltaic Multi-layer

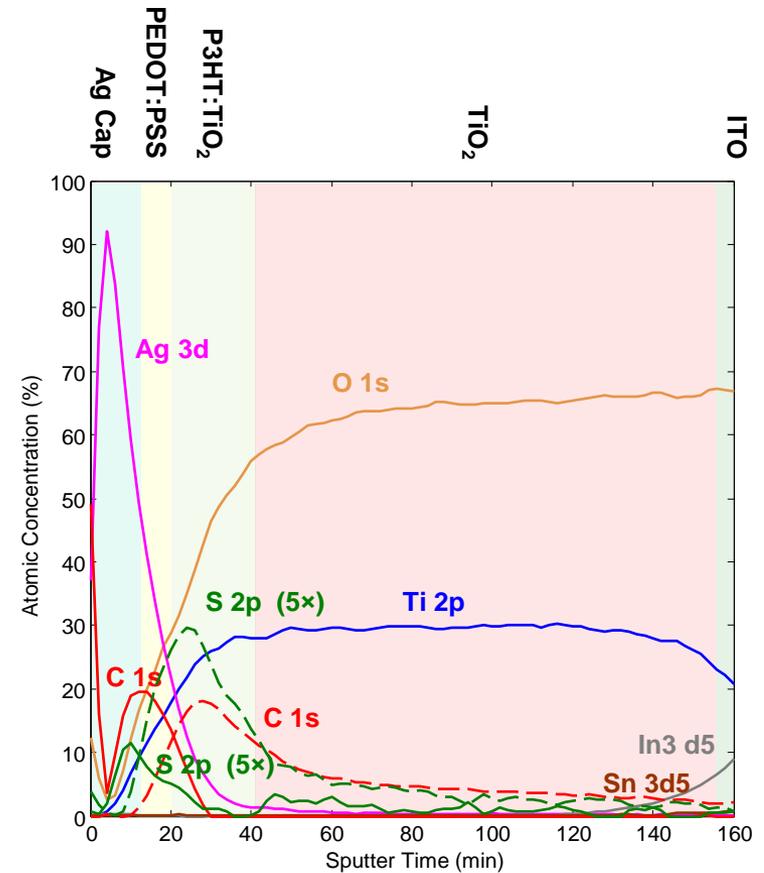
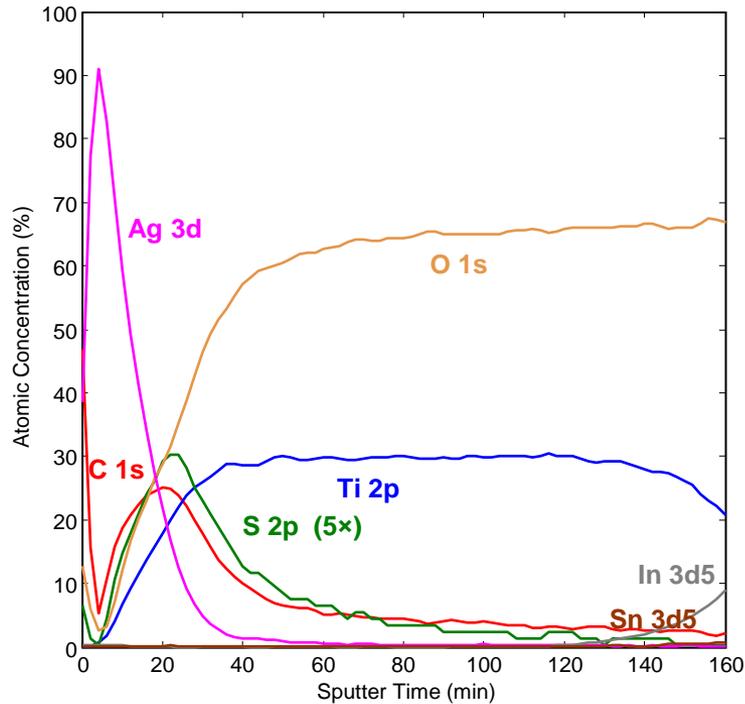
20 kV C₆₀ Cluster



- With Compucentric Zalar Rotation™
- Well defined layers
- Consistent sputter speed throughout the multilayer stack

Composite – Organic Photo Voltaic Multi-layer

20 kV Ar₁₅₀₀⁺ Gas Cluster, 13.3 eV/atom



(13 eV/atom)

- Poor layer definition
- Large variation in sputter rates

Fabrication Process versus OLED Efficiency

□ Two fabrication processes of small molecular OLED's

- Spin coating (wet process): typically easier fabrication, lower efficiency
- Evaporation has higher efficiency

□ Emissive layers (ELs) studied for high efficiency green OLED's

- Guest
- Bis[5-methyl-7-trifluoromethyl-5H-benzo©(1,5)naphthyridin-6-one]iridium (picolate) (CF₃BNO)₂IrPLA
- Host
- 4-4'-bis(carbazol-9-y)biphenyl (CBP)
- Wet efficiency 70 lm W⁻¹
- Dry efficiency 21 lm W⁻¹ WHY?

Wet versus Dry Process: XPS Depth Profiles

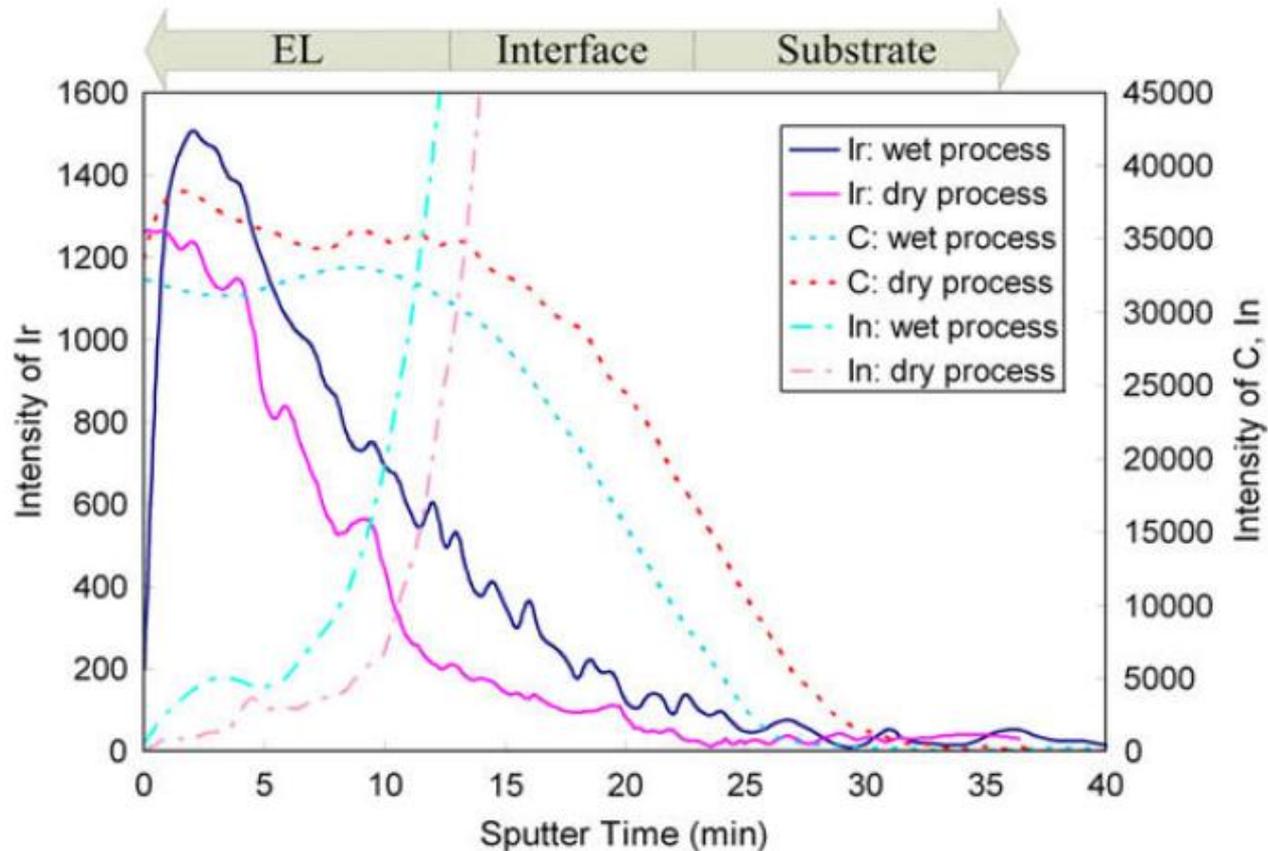


Fig. 1. XPS elemental depth profiles of emissive layers prepared with thermo-evaporation (dry-process) and spin-coating (wet-process).

- Same Ir composition similar when normalized by film thickness
- Wet process Ir guest has higher concentration at interface relative to dry process

Wet versus Dry Process: TOF-SIMS Depth Profiles

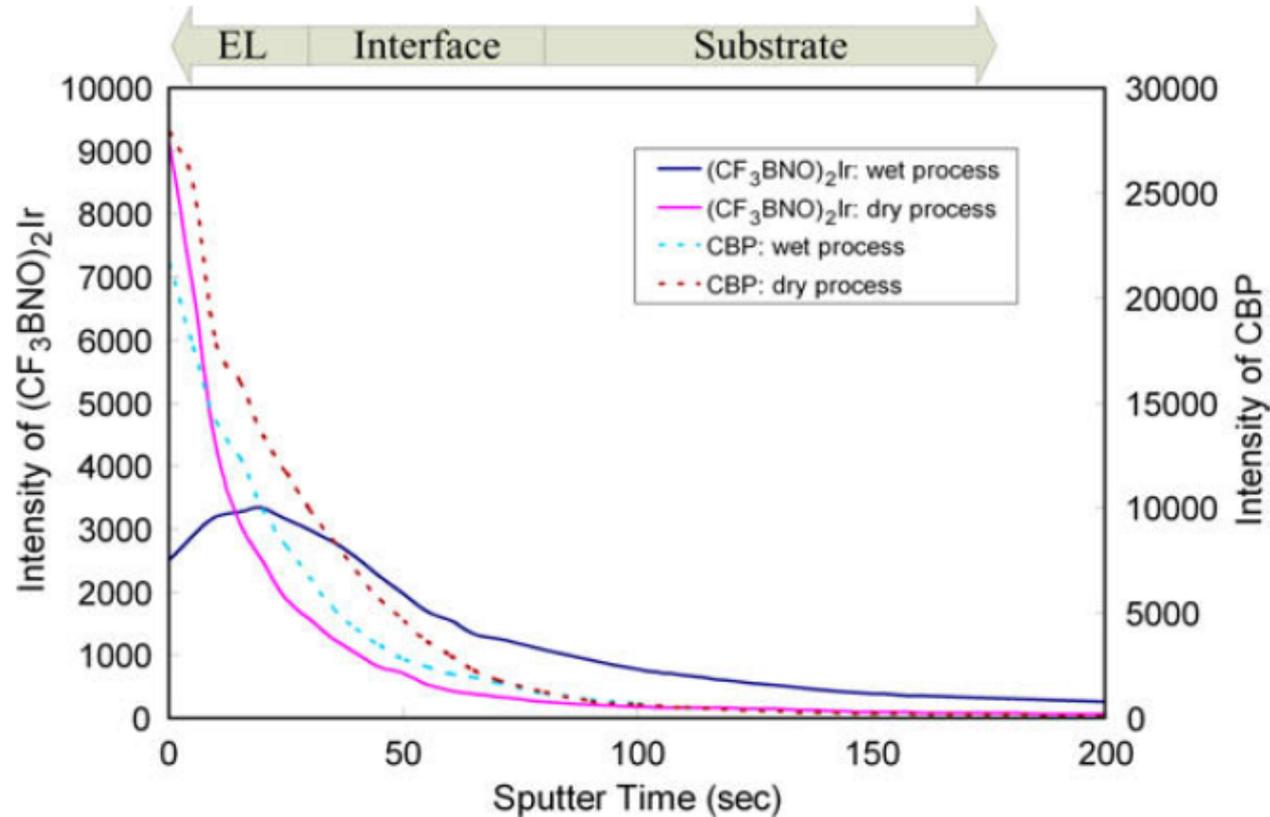


Fig. 3. ToF-SIMS depth profiles of emissive layers prepared with thermo-evaporation (dry-process) and spin-coating (wet-process).

- Same Ir composition similar when normalized by film thickness
- Wet process Ir guest has higher concentration at interface relative to dry process

Wet versus Dry Process Model of Efficiencies

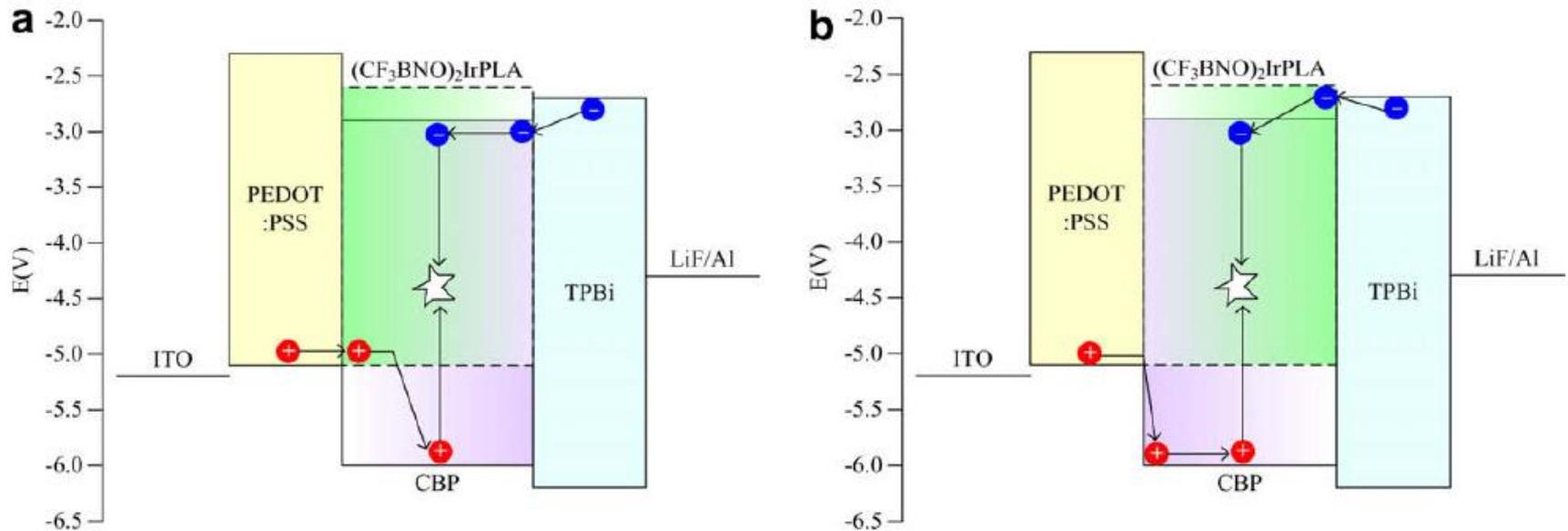


Fig. 4. Energy diagram and the route of charge carriers of OLED with EL prepared with (a) wet-process and (b) dry-process. The shade of CBP and (CF₃BNO)₂IrPLA indicates its relative concentration.

- Higher relative guest concentration at HT interface give lower turn-on voltage
- More hole trapping in dry process
- Wet process efficiency is ~ 3.5x higher than dry process

Cluster Ion Source Summary

The Choice of Ion Source for XPS Compositional Depth Profiling Is Application Dependent

Sample Type	Ion Gun Type		
	Ar Monoatomic	C ₆₀ Cluster	Ar Gas Cluster
Metal	<ul style="list-style-type: none"> Preferred approach <p>★★★★★</p>	<ul style="list-style-type: none"> Carbide formation 3 - 20 % on reactive metals <p>★★★</p>	<ul style="list-style-type: none"> Slow etch rates Small differential sputtering for some alloys <p>★</p>
Ceramics	<ul style="list-style-type: none"> Differential sputtering and chemical change <p>★</p>	<ul style="list-style-type: none"> Ideal Approach on Soda-lime glass Works well on metalloid oxides and conducting oxides: ITO / InGaO <p>★★★★★</p>	<ul style="list-style-type: none"> Slow etch rates Small oxygen loss for some oxides at high acceleration voltages (ITO) <p>★★★</p>
		<ul style="list-style-type: none"> Not suitable for transition metal oxides TiO₂ / HfO₂ / WO₃ <p>★★★</p>	<ul style="list-style-type: none"> Not suitable for transition metal oxides TiO₂ / HfO₂ / WO₃ <p>★</p>
Organics Polymers	<ul style="list-style-type: none"> Significant Chemical Damage <p>★</p>	<ul style="list-style-type: none"> Suitable for Type-II degrading polymers (5 × to 10 × between Polymers and Metal oxides) <p>★★★★★</p>	<ul style="list-style-type: none"> Suitable for Type-II degrading polymers (50 × to 100 × between Polymers and Metal oxides) <p>★★★★★</p>
		<ul style="list-style-type: none"> Some Type-I crosslinking polymer to a depth of a few hundred nanometers <p>★★★</p>	<ul style="list-style-type: none"> Ideal Approach for Type-I crosslinking polymers, unless an inorganic filler is present <p>★★★★★</p>
Complex Materials	<ul style="list-style-type: none"> Sample Dependent <p>★★★</p>	<ul style="list-style-type: none"> Able to maintain sputter speed across layers, with low level damage on Carbon <p>★★★★★</p>	<ul style="list-style-type: none"> Large variation in sputter rate causes depth resolution problems <p>★</p>
Semiconductors	<ul style="list-style-type: none"> Preferred approach <p>★★★★★</p>	<ul style="list-style-type: none"> Carbide formation on reactive metals <p>★★★</p>	<ul style="list-style-type: none"> Slow etch rates, variable sputter rates, surface roughening <p>★</p>

Summary and Conclusions

- ❑ Using Ar GCIB depth profiling and XPS we have been able to confirm the graded composition emissive layer structures
- ❑ The chemistry of the materials remains unaltered by the Ar cluster ion beam
- ❑ Sharp interfaces measured for layered structures
- ❑ The depth profile of the Ir from Ir(ppy₃) dopant is clearly observable with a concentration of < 0.04 at. %
- ❑ GCIB depth profiling is reliable reproducible and fast for these materials
- ❑ C₆₀ provides excellent depth profiling of metal/organics/metal oxides/glass multi-layer structures
- ❑ Similar use of C₆₀ and GCIB combined with TOF-SIMS can provide more molecular information