

3D Trajectories, Diffusion, and Interaction Energies of Ceria Particles on Glass Surfaces

by

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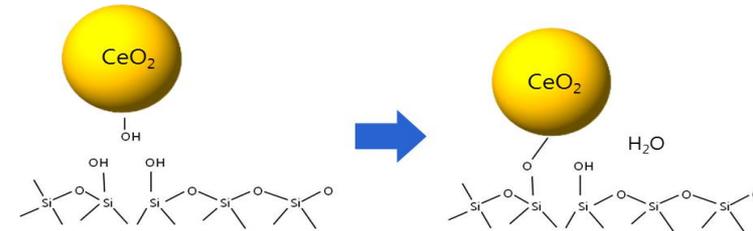
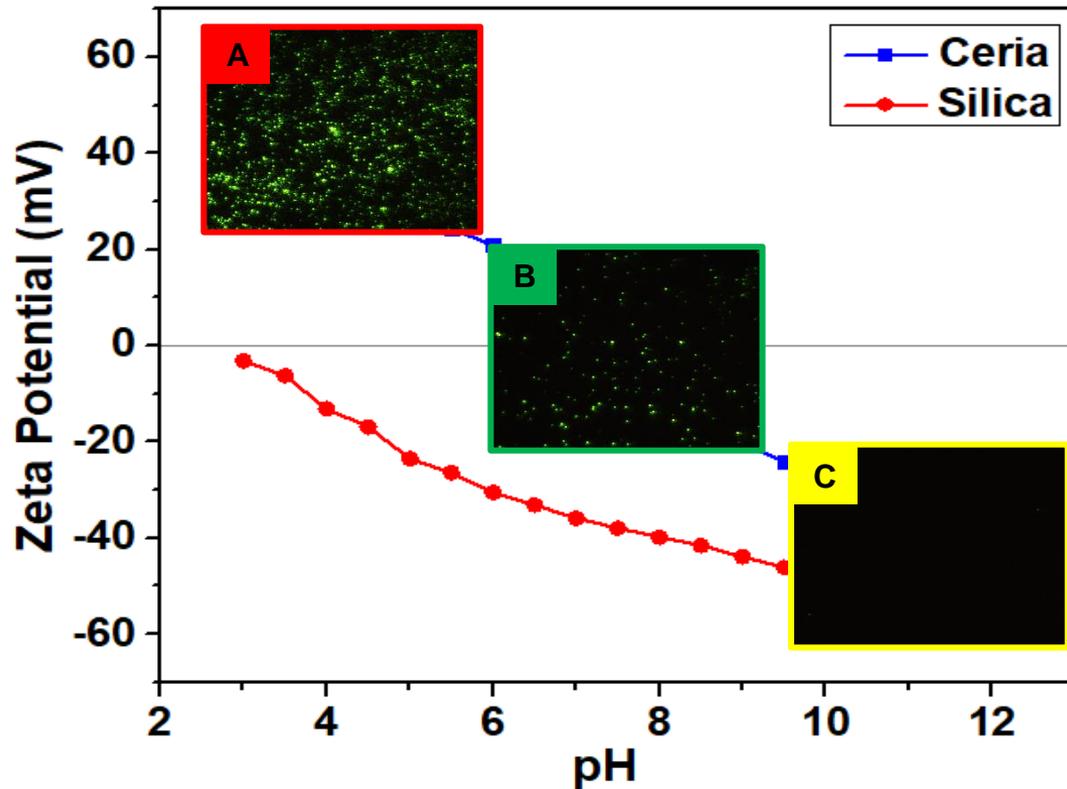
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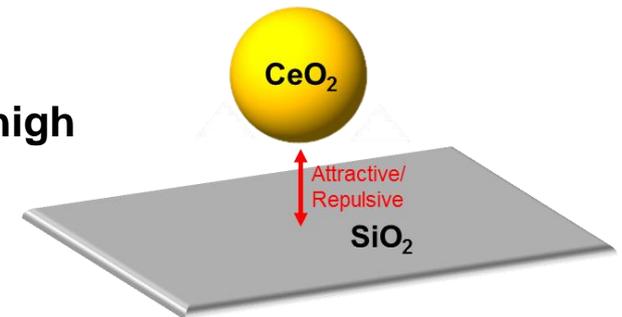
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- Introduction
- Principle of evanescent wave microscopy
- Results
 - ❖ 2D and 3D trajectories, Diffusion coefficients and Interaction potentials
 - ❖ Cleaning of ceria particles adsorbed at pH 3, pH 5 and pH 7
- Conclusions
- Acknowledgements

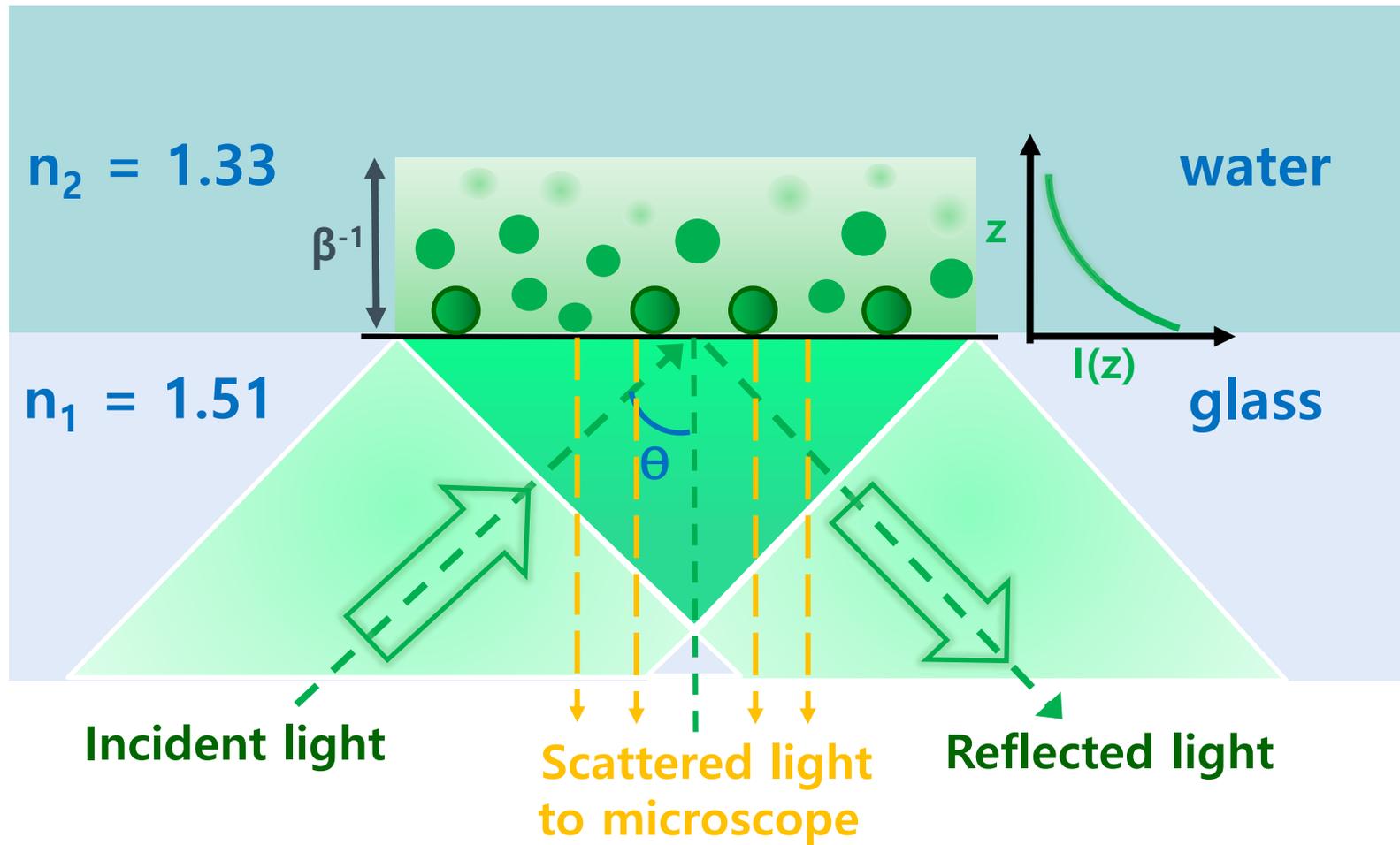
Why ceria-silica interaction ? - Motivation of this research



- ❖ Ceria based slurries in STICMP – high oxide polish rate
- ❖ Cleaning of ceria particles is challenging due to their high chemical affinity

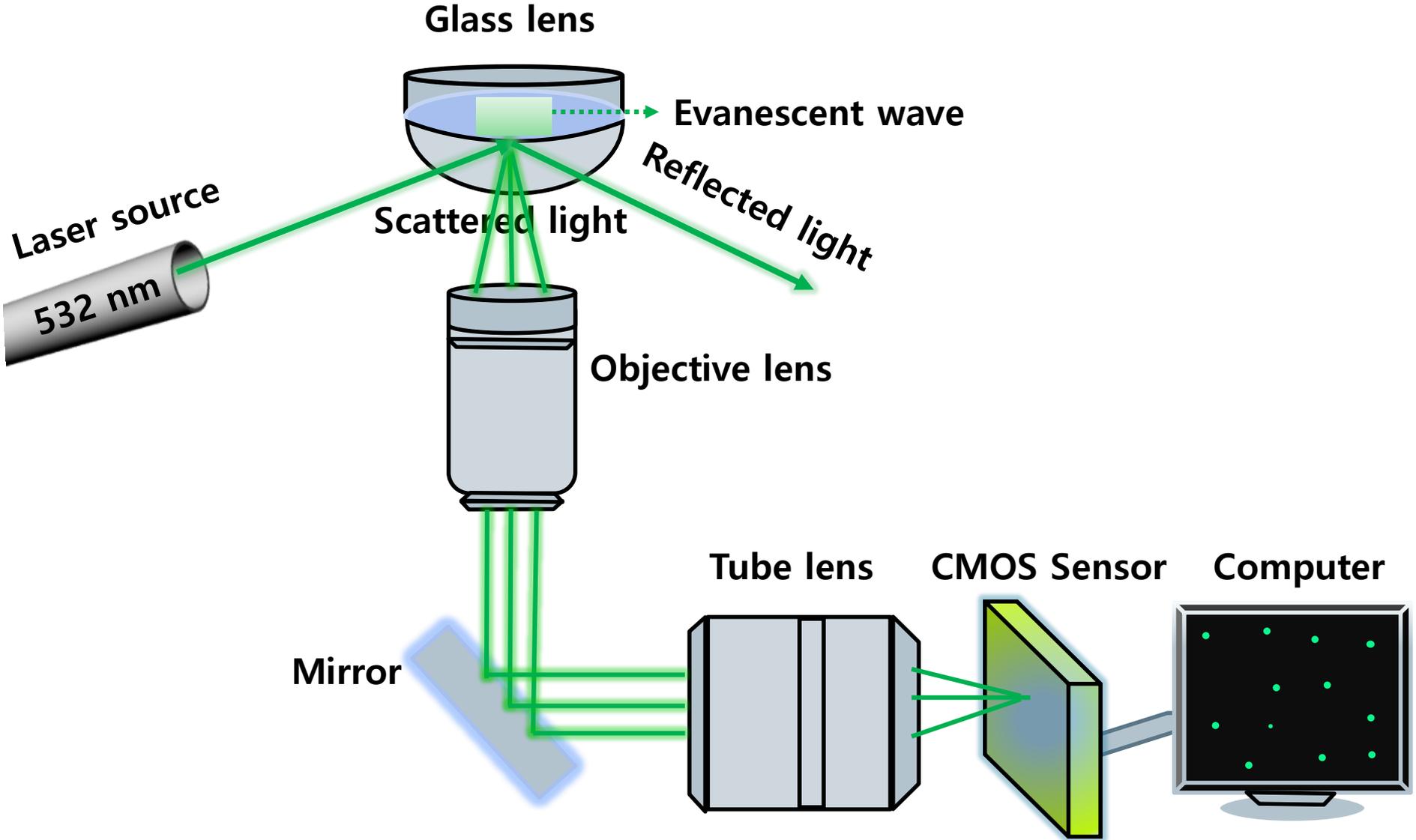


Principle of evanescent wave (EW) microscopy



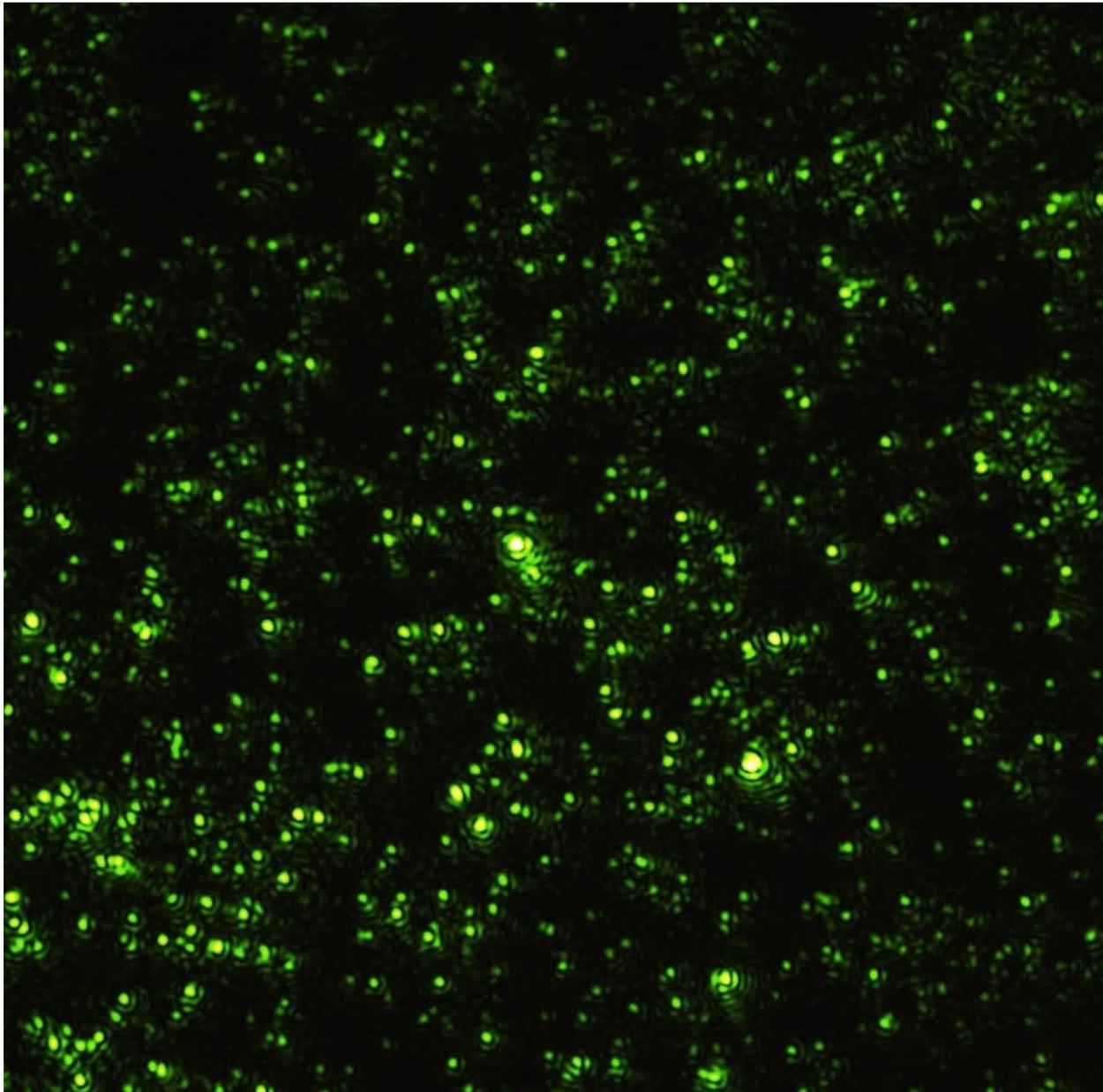
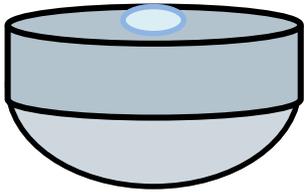
$$\beta = \frac{4\pi}{\lambda} \sqrt{(n_1 \sin \theta_i)^2 - n_2^2}$$

Experimental setup

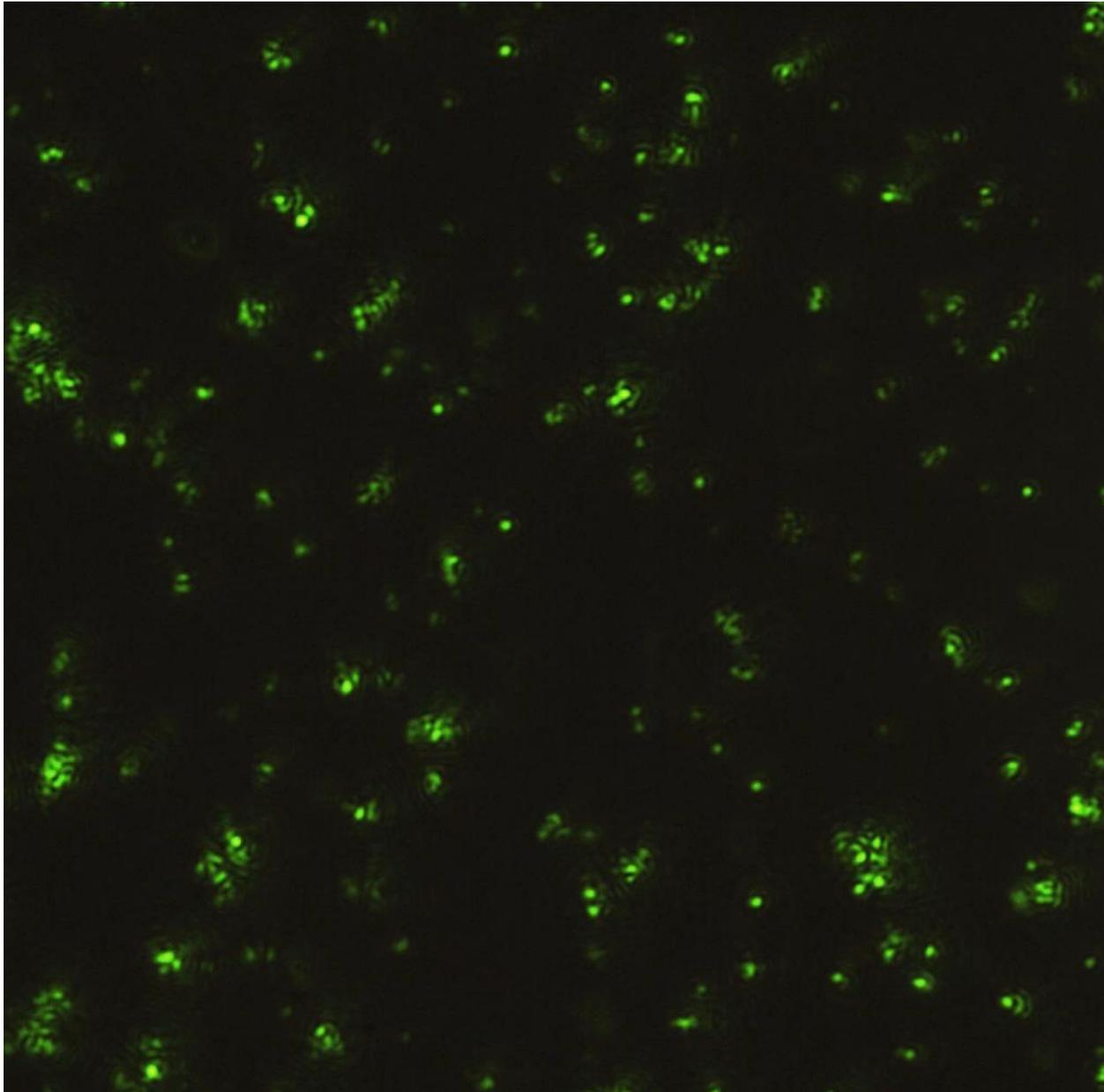


Real time imaging of ceria particles on glass film

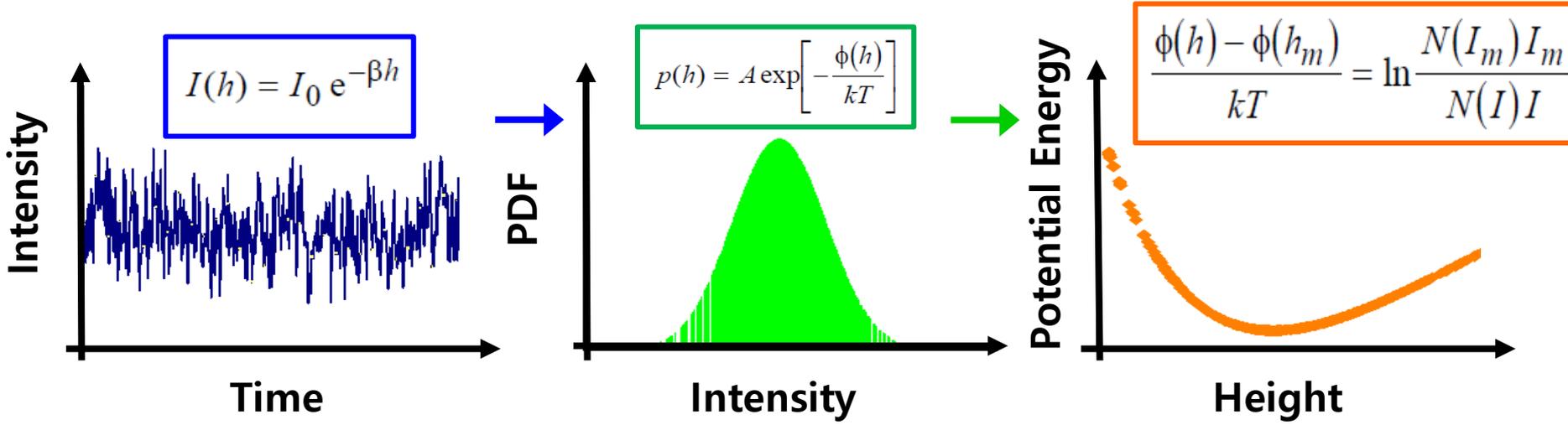
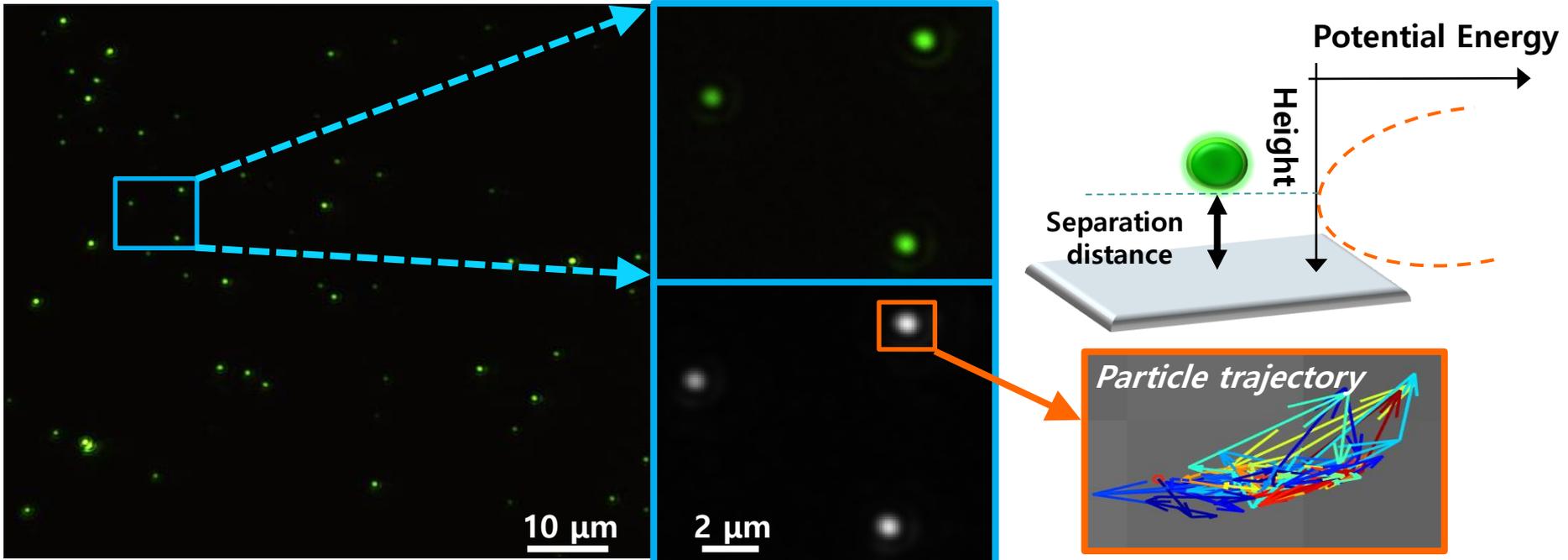
Ceria dispersion
($d_{mean} \sim 140$ nm)



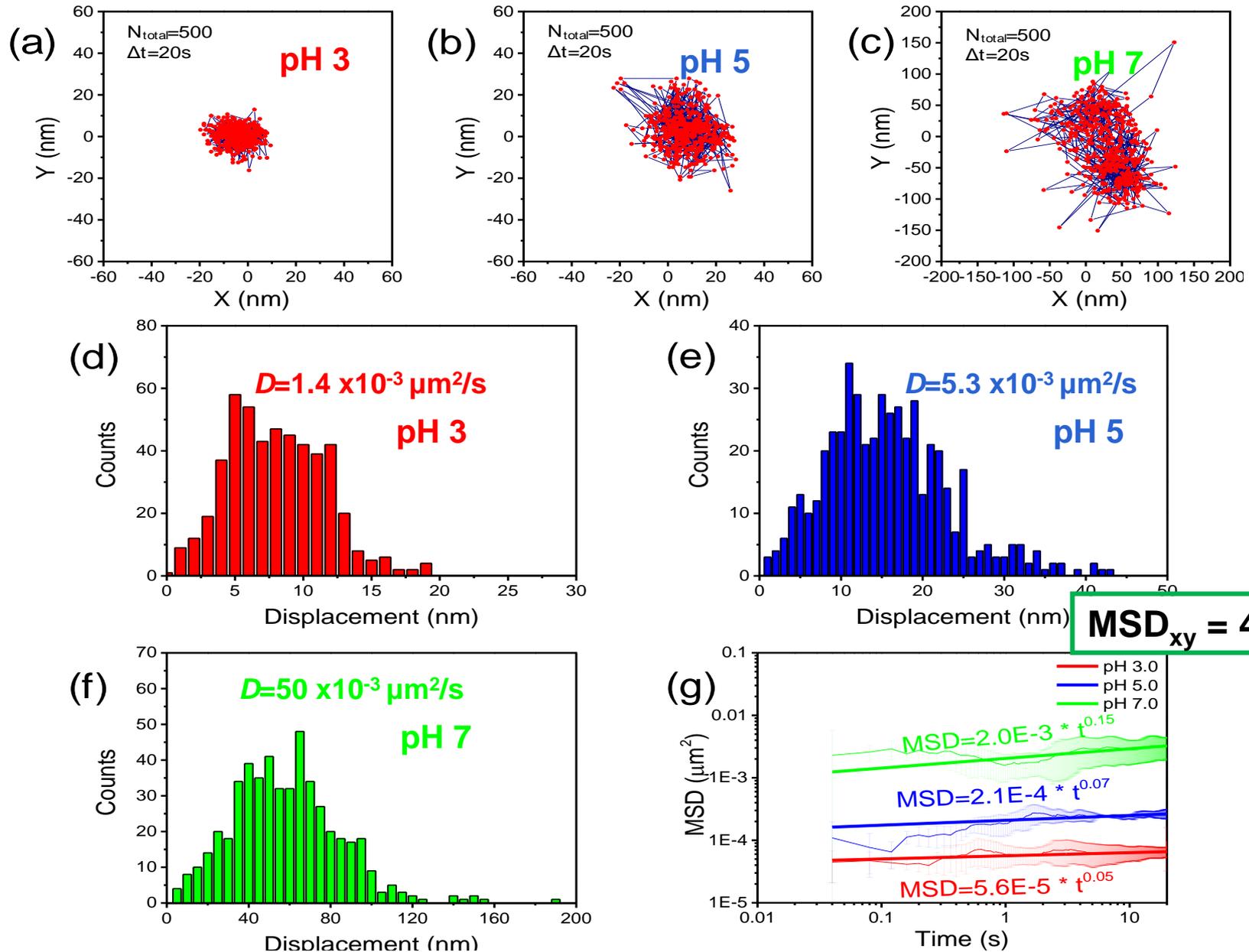
Agglomeration of ceria at pH 7



Schematic of the procedure to calculate interaction potentials



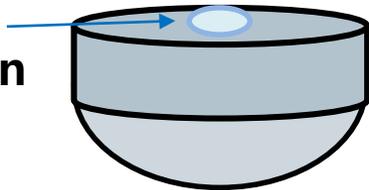
2D trajectories, histograms of particle displacements and MSD vs time



Ceria particle cleaning procedure

1. Particle adsorption

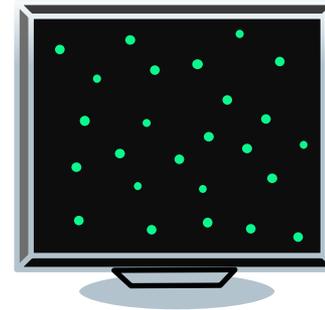
Particle dispersion



Glass lens



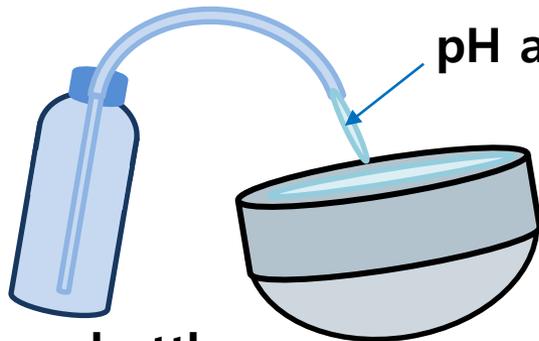
Computer



Conditions

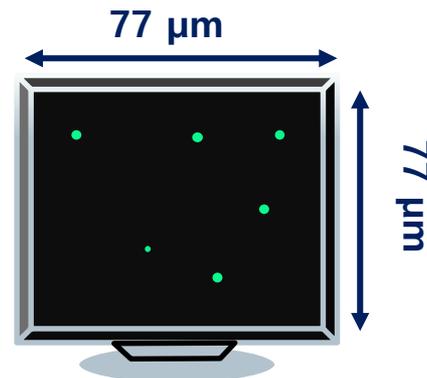
- 0.005 wt% Ferro ceria dispersion ($d_{\text{mean}} \sim 140 \text{ nm}$)
- Particles adsorbed for 20 minutes

2. Particle cleaning



pH adjusted DI water

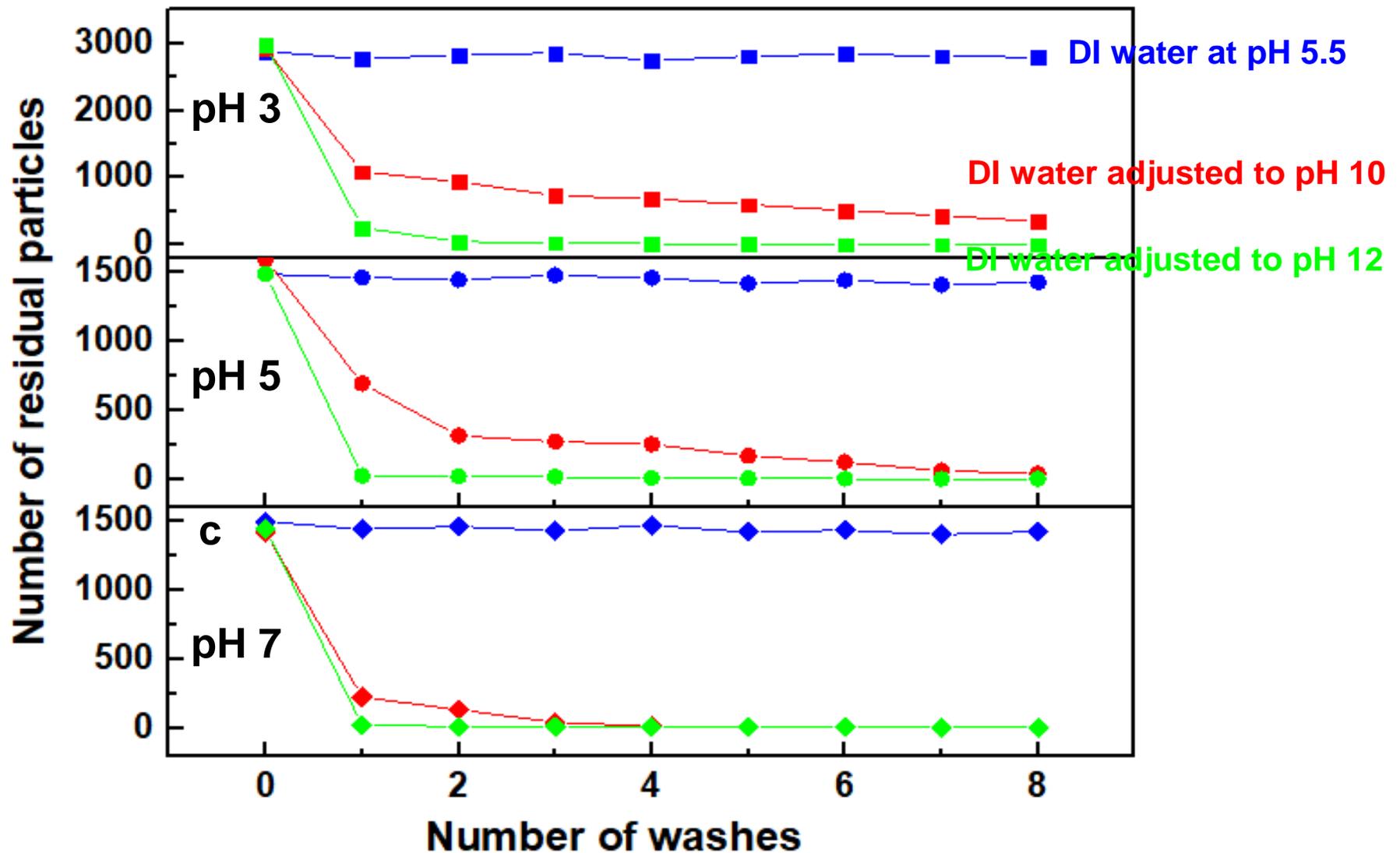
Squeeze bottle



Procedure

- Lens covered with ceria particles was washed multiple times.
- Each time with 40 mL of pH adjusted DI water for 20s

Number of residual particles vs number of washes



Conclusions

1. Evanescent wave microscopy is a powerful tool to study particle-surface interactions.
2. Ceria particles at pH 5.0 ($D_{3D} = 3.8 \times 10^{-3} \mu\text{m}^2/\text{s}$) and pH 7.0 ($D_{3D} = 45 \times 10^{-3} \mu\text{m}^2/\text{s}$) exhibited faster diffusion near glass surface than at pH 3.0 ($D_{3D} = 1.2 \times 10^{-3} \mu\text{m}^2/\text{s}$)
3. As pH increased from 3.0 to 7.0, the most probable separation distance h_{min} of ceria particle from glass surface increased from 11 to 60 nm.
4. Ceria particles adsorbed at higher pH can be more easily cleaned.
5. Charge repulsion is shown to be a key driver towards cleaning performance.
6. While charge repulsion is enough to clean larger ceria particles, it is not efficient in removing smaller particles

THANK YOU!
